

# MIND MAPS IN CLINICAL CHEMISTRY

**PART I**



**Simmi Kharb**

**Bentham Books**

# **Mind Maps in Clinical Chemistry**

*(Part I)*

Authored by

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## **Mind Maps in Clinical Chemistry (Part I)**

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## **PREFACE**

This book treats clinical chemistry as an interdisciplinary subject to teach the importance of lab both for students in basic science as well as clinical medicine.

The approach to clinical chemistry in real clinical situations requires knowledge, experience and an integrated and clinically relevant model. I have attempted to bring this integrated model together in the book.

The book has been written in a lucid manner and is packed with practical approach of lab tests targeting undergraduate and also 1st postgraduate students of various streams of medicine. The reader should gain the knowledge and understanding of the value, limitations and interpretation of lab tests in modern medicine.

### **ACKNOWLEDGMENT & CONFLICT OF INTEREST**

No potential conflict of interest is declared by the author. It is also declared the complete work is an individual effort by the author and there was no financial/administrative/academic support availed from any individual/institution/organization.

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# Section I

## Introduction



# Collection and Processing of Specimens

## Learning objectives:

1. Describe the method of sample collection.
2. Differentiate the factors affecting lab results of the patient.
3. Explain about sample processing.
4. Recognize and analyse the causes of errors in the lab.
5. Define POCT (Point of Care Testing) and its applications.

## Collection of Sample

While blood, serum, plasma, urine, feces, saliva, sweat, amniotic, cerebrospinal, pericardial, peritoneal, pleural and synovial fluids are the biological specimens, they are analyzed in clinical chemistry laboratories.

- Steps -
1. Preanalytical testing
    - Request
    - Preparation of patient
    - Preparation for collecting sample
  2. Sample collection

## Lab Test Request

The process of clinical laboratory investigation starts with a clinician making a request.

Test request form includes patient's demographics:

- patient's name, age, sex
- date of admission
- date on which request for lab test is made
- ward/clinic/address
- clinical diagnosis
- tests requested
- physician's identity.

Test request information is conveyed through written or computer order entry.

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## Patient

- Correct interpretation of lab results can only be performed after careful consideration of individual patient factors (Table 1.1).

**Table 1.1:** Patient factors affecting interpretation of lab results

Age
Sex
Stress
Body temperature
Previous intake of food or liquid
Drug therapy
Race, individual variables

In general, patients scheduled for sampling should refrain from strenuous physical activity, alcohol, drugs, dietary change and should have usual time and amount of sleep (Table 1.2).

**Table 1.2:** Patient factors affecting lab reports

Factor	Examples	Reasoning
1. <i>Between individual differences</i>		
Age	urea, alkaline phosphatase	Rise with age
Sex	uric acid	Higher in males
	gonadal steroids	Different between sexes
	HDL-C	Higher in premenopausal women
Race	Creatine kinase	Higher in Africans than Caucasians
2. <i>Within individual variations</i>		
Circadian rhythm*	Cortisol	Highest between 7-9 am Lowest between 11 pm to 4 a m Rise in stress Lost in Cushing's syndrome Higher in morning Reversed in night workers Monthly variation in female Low concentration just before or during menses
Iron		
Diurnal rhythm		
*	Travel across several time zones affects normal circadian rhythm. Also, during prolonged flight, fluid and sodium retention occurs.	
Others		
3. Pregnancy	Thyroxine	Thyroxine binding globulin elevate from supine to upright position
Posture	Proteins	upright position elevated
Exercise	Creatine kinase	
Nutritional status:		
High meat or protein:	Urea, urate,	Rises

Contd...

Contd...

serotonin-rich diet: bananas, pine apple, tomato		raised serotonin
recent food intake: FF, glucose, lipid		increased
fasting	Glucose	Fall
time	Bilirubin	Rise after 48 hrs of fasting
	Cortisol	Circadian rhythm
	Uric acid	Elevated
Alcohol	Lactic acid	Elevated
	Triglyceride	Elevated
	GGT	Elevated
	Carboxyhemoglobin	High levels
Smoking	Cortisol	High levels
	Catecholamine	High levels
	Protein, AST	Decrease
Immobilization		Excretion increased
		Falsely increased
Tourniquet	Ca, Na, K, PO <sub>4</sub> , SO <sub>4</sub>	
	Lactate, calcium, proteins	
	triglycerides	
Stress	Hormone,	Hormone secretion affected
	Lactate, FFA	Elevated

## Few Important Points

### How Often a Test should be Ordered?

- i. Over investigation may be harmful, causing unnecessary discomfort or inconvenience to patient.
- ii. Frequency of sampling should be carefully decided, e.g. plasma protein concentration are unlikely to change significantly in less than a week, plasma urea will not change significantly during first 12 hours of oliguria.
- iii. Numerically significant changes sometimes do not alter treatment or course of disease, e.g. plasma transaminases.
- iv. Plasma potassium concentration may alter after diuretics and indicate need to investigate or change treatment.

## Urgency

An urgent lab request is indicated when that investigation will alter the clinical management of patient.

## Are the Differences Significant?

Biochemical results obtained on specimens taken sequentially from a single individual are frequently numerically different; consider the possible sources of variation:

- i. Pre-analytical variation

# Lab Hazards:

## Biological Hazards

### Learning objectives:

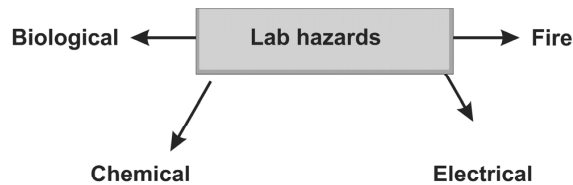
1. Recognise the different lab hazards.
2. Illustrate Labelling identification system.

To operate a clinical laboratory safely, it is essential to prevent the exposure of laboratory workers to infectious agents (Hepatitis B virus, HIV) by:

- i. accidental puncture with needles
- ii. spraying of these materials by syringe or spilling on benchtop or floor
- iii. centrifuge accidents
- iv. cuts or scratches from contaminated vessels (Fig 2.1)

It is essential that specimens should be handled using universal precautions such as:

1. Handle all specimens with care by using barriers protection (glove, gown, lab coats, face shield, eye protection)
2. Never mouth pipette or blow out pipettes that contain potentially infectious materials.
3. Dispose all sharps appropriately.
4. Hepatitis B vaccination should be offered to all employees at risk of potential exposure (technician, phlebotomist, etc.) (Fig. 2.1).



**Fig. 2.1:** Lab hazards

Nine classes of hazardous materials (UN classification)

- |                             |                        |
|-----------------------------|------------------------|
| 1. Explosives               | 2. Compressed gases    |
| 3. Flammable liquids        | 4. Flammable solids    |
| 5. Oxidizer materials       | 6. Toxic materials     |
| 7. Radioactive materials    | 8. Corrosive materials |
| 9. Miscellaneous materials. |                        |

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### Hazard symbols and Labelling

Hazard symbols are recognizable symbols to warn about hazardous materials or locations. They may appear with different colors, backgrounds, border and supplemental information in order to signify the hazard (Fig 2.2, Fig 2.3)

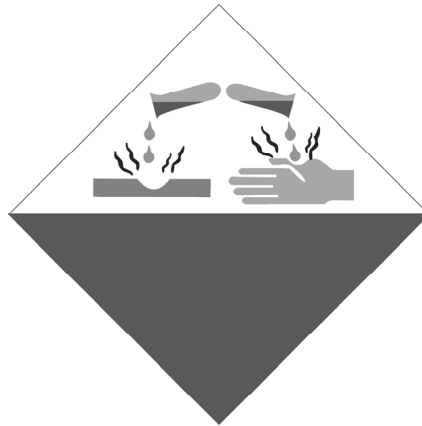


Fig. 2.2: Label for corrosive

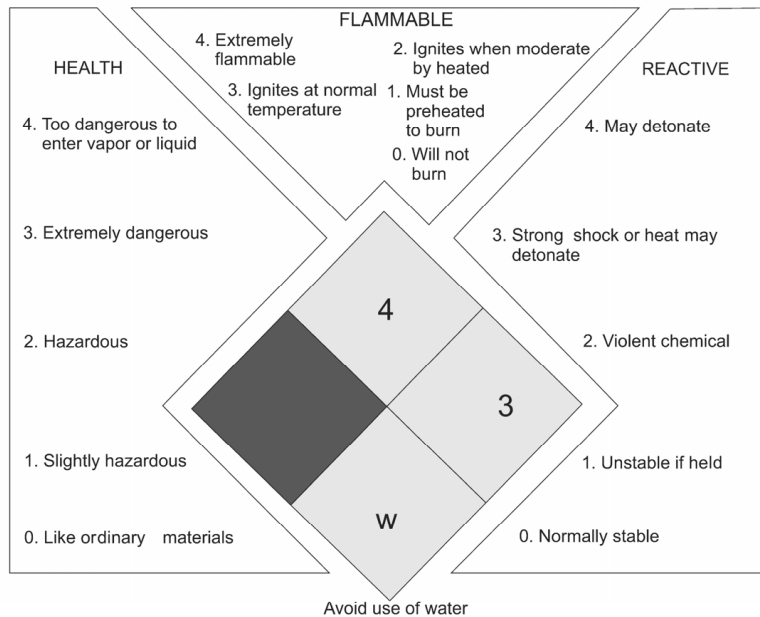


Fig. 2.2: Labeling identification system

Fig. 2.3: Labeling identification system

## **Safety in the Laboratory**

One must always be aware that chemicals used in the laboratory are potentially toxic, irritating or inflammable. Such chemicals are a hazard, however, only when they are mishandled or improperly disposed off.

Proper disposal of all waste chemicals, sharp objects, and infectious agents is essentially not only to maintain safe laboratory working condition, but also to protect the general public and local environment.

### **Questions:**

1. What are biological lab hazards?
2. Classify hazardous materials?
3. What are hazard symbols and labelling?
4. Elaborate the various safety measures taken in laboratory?
5. Explain Labeling identification system.

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# Safety in the Laboratory

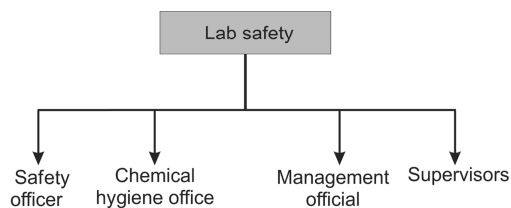
## Learning objectives:

1. Evaluate the aspects of safe operations of a Clinical Chemistry lab.
2. Explain the guidelines for Lab safety.

A clinical chemistry laboratory should have formal safety program.

Aspects of Safe Operations of a Clinical Chemistry Lab:

- Formal safety program
- Mandated plans:
  - Chemical hygiene plans
  - Blood-borne pathogen plans
- Identification of lab hazards:
  - biological
  - chemical
  - electrical
  - fire
- OSHA, CDC approved hazard
- Communication standards



**Fig. 3.1:** Lab safety network material

## Guidelines for Lab Safety

1. Each new employee should get general lab safety material, undergo orientation program.
2. Each employee should have knowledge of location of available evacuation routes, firefighting equipments and their operation.
3. Employee should undergo continuing education programs on lab safety.

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4. Supervision of lab safety measures in terms of:
  - proper labeling, handling and disposal of chemicals in terms of hazards
  - fire extinguisher
  - hoods
  - earthing of electrical equipment
  - proper handling and disposal of patient specimens.
5. Inspection: Periodic and scheduled safety inspections should be the routine.
6. Accreditation of lab with Health Care Organization.
7. Safe and scientific waste disposal and management.

**Questions:**

1. What are the aspects of Safe Operations of a Clinical Chemistry Lab?
2. Write the guidelines for lab safety.

**Bibliography:**

2004. Laboratory Biosafety Manual. Geneva: World Health Organization.
2011. Prudent Practices In The Laboratory. Washington, D.C.: National Academies Press.
- Belloc, H., 1967. On. Freeport, N.Y.: Books for Libraries Press.
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- Tietz, N., 1987. Fundamentals Of Clinical Chemistry. Philadelphia: W.B. Saunders.



# Biochemical Tests in Clinical Medicine

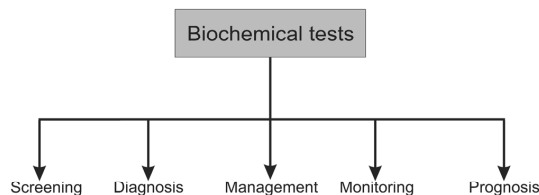
## Learning objectives:

1. Discuss the purpose of biochemical tests.
2. Illustrate the outline of Lab workflow.

The purpose and function of laboratorians are to provide biochemical information to assist clinicians (Fig. 4.1) in:

1. diagnosis
2. management
3. prognosis
4. detection and screening
5. monitoring follow-up of disease
6. teaching, research and development.

For results to be useful for any of these purposes, they must be free of error.



**Fig. 4.1:** Purpose of lab tests

## Lab Test Request

Most clinicians use lab tests primarily on the basis of their own clinical expertise and interpret results intuitively, and their results are used on the basis of outcome measure. In selecting a test, it is essential to consider what type of information is required and whether the test would provide it or not.

A clinical chemistry laboratory performs a large number of biochemical analytes on body fluids which can give answers to specific clinical questions about an individual patient. Such analyses are requested by clinicians in blood and urine samples or treatment.

The entire process from a request for analysis to receipt of a result involves several steps and can take many hours. Also, quality assurance for analyte is performed to ensure that results thus produced are analytically and clinically valid. Also timeliness, promptness or turnaround time of a test report is equally important in patient care.

The usual procedure by which a clinician obtains a medically useful result is an extremely complex chain of events (Fig. 1.1). This chain involves: clinicians, nurses, porter, technical staff, clinical biochemists or lab physician. Errors can occur at every link in the chain.

Samples to be analyzed must be collected under appropriate conditions and the analytical methods selected must be reliable. Also, the possible contribution of any analytical or biological variation should be considered.

## Outline of Lab Workflow

---

Steps involved in generation of lab results (Fig. 1.4) towards a test result must be considered. The laboratory personnel must ensure that data are accurate and precise and the clinician should appreciate how useful the test is, the context in which it is used. Also, clinical staff should avail the opportunity to consult their lab colleagues about applications of lab tests in management of their patients.

### Questions:

1. What is the purpose and functions of lab tests?
2. Discuss Lab Test Request.

### Bibliography:

- Burtis, C., Ashwood, E., Bruns, D. and Tietz, N., 2008. *Tietz Fundamentals Of Clinical Chemistry*. St. Louis, Mo.: Saunders Elsevier.
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- Luxon, S., 1992. *Hazards In The Chemical Laboratory*. Cambridge: Royal Society of Chemistry.

## Section II

# Metabolic Disorders

**CHAPTER 5**

# Bone Metabolism: Calcium, Phosphate and Magnesium

**Learning objectives:**

1. Discuss the normal metabolism of Calcium, Phosphorus and Magnesium in body.
2. Describe the metabolic disorders associated with Calcium, Phosphorus and Magnesium.
3. Demonstrate the markers of Bone Turnover.
4. Compare different disorders of mineralization of bone.

**Normal Calcium Homeostasis**

Calcium is the most abundant mineral in the human body. Bones and teeth contain 99 percent of calcium. An average human body contains 1 kg (24.95 mol) of calcium (Fig 5.1). Calcium is present in three main compartments:

- Skeleton 99 percent
- Soft tissue
- Extracellular fluid (ecf) 1 percent

Calcium is physiologically classified as either intracellular or extracellular. The skeleton is major reservoir for providing both extracellular and intracellular pools.

Intracellular calcium	Physiology of calcium Muscle contraction Hormonal secretion Glycogen metabolism Cell division
Extracellular calcium	Source for maintenance of intracellular calcium for bone mineralization Coagulation cascade Maintaining plasma membrane potential

**Functions of calcium**

Structural	Bone Teeth Controls membrane potential
Neuromuscular	Neuromuscular excitability Neurotransmitter release
Enzymatic Signaling	Coagulation cascade (as coenzyme) Intracellular second messenger

Total calcium in extracellular fluid is only 22.5 mmol (10 mg/dl) of which 9 mmol (4.1 mg/dL) is in plasma.

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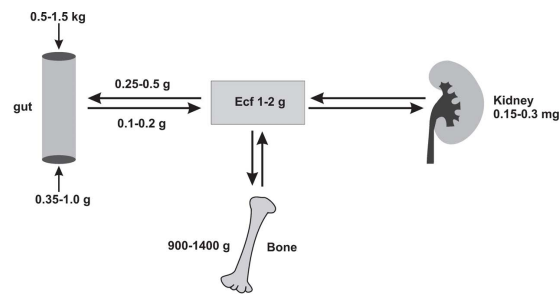


Fig 5.1: Calcium homeostasis

## Calcium in Plasma

Calcium exists in three physiochemical states in plasma:

- 50 percent free or ionized
- 40 percent bound to plasma proteins: 80 percent with albumin, 20 percent with globulin
- 10 percent complexed with small anions.

## Calcium Concentration

In normal adults, range of plasma concentration is 2.2 to 2.6 mmol/L (8.8 to 10.4 mg/dL). The concentration of free calcium ions, mean value being 1.2 mmol/L (4 to 8 mg/dL), is subject to tight hormonal control (Fig. 5.2). A decrease in serum free calcium concentration causes increased neuromuscular excitability and tetany.

## Calcium Intake and Absorption

### Calcium Intake

Average dietary calcium intake should be 0.6 to 0.8 g/d in adults. Less than half of it is absorbed. (Fig. 5.2)

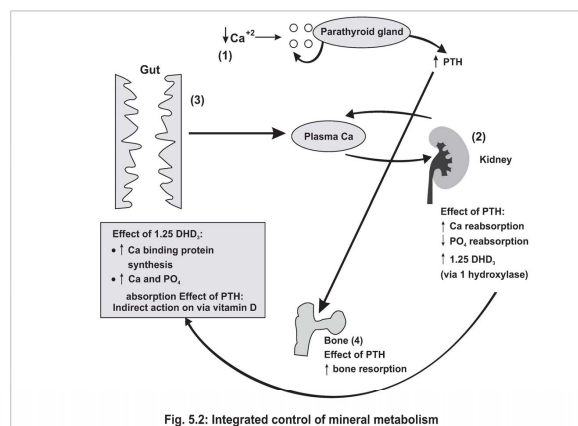


Fig. 5.2: Integrated control of mineral metabolism

## Calcium Absorption

- Increased during period of rapid growth in - children  
 - pregnancy  
 - lactation
- Decreased with advancing age

## Site

Most calcium absorbed in proximal small intestine by active transport in upper part and by simple diffusion in lower part.

**Table 5.1:** Factors affecting absorption

	<i>Favor</i>	<i>Decrease</i>
<i>pH</i>	Acidic pH	Alkaline pH
<i>Dietary factors:</i>	High protein-diet (esp. lysine, arginine) sugar and organic acid citric acid	Oxalate, phytate fibers Excess PO <sub>4</sub> and Mg Fe

## Calcium Homeostasis

### Key Target Organs of Control

Intestine  
 Parathyroid glands  
 Kidneys

# Carbohydrates

## Learning objectives:

1. Describe the role of different enzymes and hormones involved in carbohydrate metabolism.
2. Analyse the mechanism of blood glucose homeostasis.
3. Discuss Inborn errors of metabolism.
4. Describe disorders of Carbohydrates metabolism in detail.
5. Differentiate between Type I and Type II Diabetes Mellitus.
6. Explain the metabolic complications of Diabetes.

Carbohydrates are widely distributed in plants and animals performing multiple functions (Table 6.1).

**Table 6.1:** Functions of carbohydrates

- |   |
|---|
| <ol style="list-style-type: none"><li>1. Structural</li><li>2. Source of energy</li><li>3. Storage</li><li>4. Constituent of:<ul style="list-style-type: none"><li>DNA and RNA (deoxyribose, ribose sugars)</li><li>Proteoglycan</li><li>Glycosaminoglycan</li><li>Glycoproteins</li><li>Cell membranes</li></ul></li><li>5. Stored as glycogen</li></ol> |
|---|

Glucose is a major fuel of tissues and cells. The body depends on glucose for energy. Body closely regulates blood glucose to maintain an adequate supply of glucose for cells. Diseases associated with carbohydrates include diabetes mellitus, galactosemia, glycogen storage diseases, and lactose intolerance. This chapter discusses carbohydrate metabolism and its abnormalities.

## Metabolism: Carbohydrates

Through the process of digestion and absorption, carbohydrates in diets following metabolism provide glucose — primary source of energy. (Figs 6.1 to 6.3 and Table 6.1).

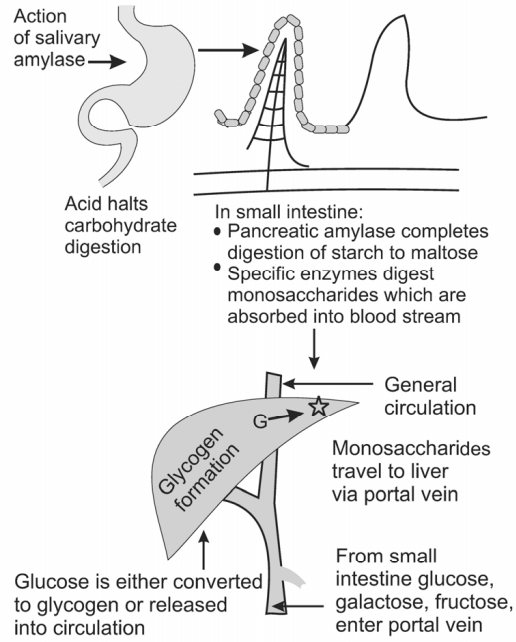


Fig. 6.1: Digestion and absorption of carbohydrates

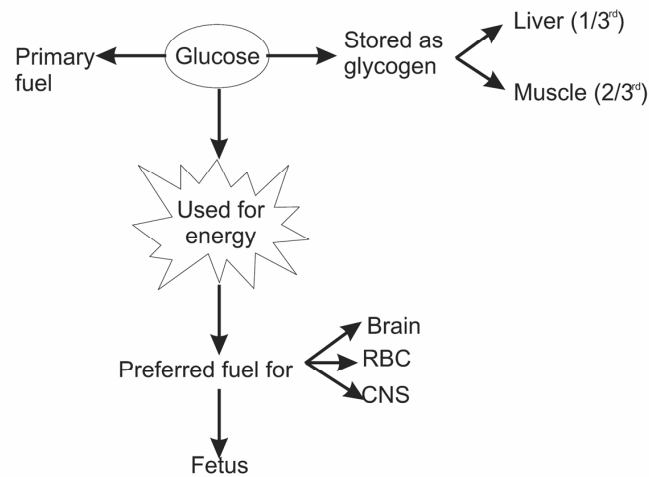


Fig. 6.2: Glucose in action

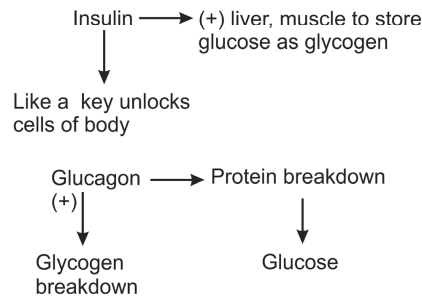
## REGULATION OF BLOOD GLUCOSE

Main hormones regulating blood glucose:

Insulin  
Glucagon

(Fig 6.3, Fig 6.4, Table 6.2)





**Fig. 6.3:** Actions of insulin and glucagon

**Table 6.2:** Hormones affecting carbohydrate, lipid metabolism and protein metabolism

Enzymes	Fed state	Fasting	Insulin	Glucagon	Other activators/ inhibitions
I					
1. glycogen synthase	+	-	(+)	(-)	G-6-P
II. GLYCOLYSIS					
1. Hexokinase			(+)	(-)	AMP, F-6-P, F2,6-BIS P/ CITRATE, ATP
2. Glucokinase	+	-	(+)	(-)	F1,6 BIS P/ ATP, ALANINE
3. PFK-1	+	-	(+)	(-)	ADD:ADP,NAD,PYRUVAT, CoA/ATP, NADH, ACETYL CoA
4. Pyruvate kinase	+	-	(+)	(-)	
5. PDH	+	-	(+)	(+)	
GLUCONEOGENESIS					
III					
1. Pyruvate carboxylose	-	+	-	+	Acetyl CoA/ADP
2. PEP carboxykinase	-	+	-	+	(-)
3. F-1-6, bisphosphatase	-	+	-	+	-/F 1, 6 bis PO <sub>4</sub> , AMP, F 2, 6 bis PO <sub>4</sub>
4. G-6-Pase	-	+	-	+	
IV					
1. G-6-PD			+		
2. 6- DeH	+		+		
V					
LIPID METABOLISM					
LIPOGENESIS			+		-
1. Acetyl CoA carboxylase	+	-	+		
2. FA synthase	+	-	+		
VI					
LIPOLYSIS					
VII					
PROTEIN					
Synthesis			(-)		
Breakdown			(-)		

Net effect:

Insulin: Plasma glucose and NEFA fall

Glucagon: Plasma glucose rise

# Lipids and Lipoproteins

## Learning objectives:

1. Define the role of Fatty Acids in lipid formation and their classification.
2. Illustrate different groups of Lipoproteins.
3. Discuss about disorders of Lipid metabolism.

Lipids play an important role in almost all aspects of biological life (Table 7.1).

**Table 7.1:** Functions of lipid

Structural component in cell
Structural and functional component in biomembranes
Provides insulation
Serves as hormone or hormone precursors
Aids in digestion
Involved in metabolic and hormonal pathway
Source of energy

Knowledge of lipid disorders is important since they are common in clinical practice and have a role in atherosclerosis.

### NEFA (Non-esterified FA)

They are liberated from adipose tissue by action of lipase, transported to liver and muscle, bound to albumin, NEFA are important source of energy in body.

## Essential Fatty Acid

Such FA which can not be synthesized in the body and must be supplied in diet.

e.g. Linoleic acid (C18 :  $\Delta^{9,12}$ ) ( Table 7.2).

**Table 7.2:** Major FA in plasma

Carbon	Saturated	Unsaturated	Source
C16	Palmitic acid	Palmitoleic Acid	Plant
	(C16 : 0)	(C16 $\Delta^9$ )	
C18	Stearic acid (C18 : 0)	Linoleic Acid (C18 : $\Delta^{9,12}$ )	Animal/Plant oil
		Linoleic Acid (C18 : $\Delta^{9,12,15}$ )	Plant
		Arachidonic Acid (C20 : $\Delta^{5,8,16,14}$ )	–
		Eicosapentanoic Acid (C20)	Fish oil

## Lipids: Classification

---

### Based on Chemical Structure

#### Fatty Acids

Short chain FA	(C <sub>2-4</sub> )
Medium FA	(C <sub>6-10</sub> )
Long chain FA	(C <sub>12-26</sub> )
Eicosanoids	(C <sub>20</sub> )

#### Glycerols

TA G  
DAG  
MAG  
Phosphoglycerides

#### Sphingosine Derivatives

Sphingomyelin  
Glycosphingolipids

#### Sterol Derivatives

Cholesterol  
Vitamin D  
Steroid hormones

#### Isoprenes

Vitamin A  
Vitamin E  
Vitamin K

### PPAR: Peroxisome Proliferator- Activator (Table 7.3)

**Table 7.3:** PPAR: Peroxisome proliferator - activator receptors

<p>Family of nuclear receptors Activated by FA Implicated in insulin resistance, dyslipidemia Types : <math>\alpha</math> PPAR <math>\gamma</math> PPAR Drugs targets for different PPAR <math>\alpha</math> (+) by fibrates <math>\gamma</math> (+) by thiazolidenedione</p> <p>e.g. Pioglitazone Rosiglitazone</p>
--

## TAG (Triacylglycerol): Also called as triglycerides (TG)

Storage form of fat

Transported from intestine to liver and adipose tissue as lipoproteins

Contain both saturated and unsaturated FA.

### Cholesterol: Salient Features

Present in all cells and body fluids  
 Precursor of steroid hormones and bile acids  
 Absorbed from intestine and also synthesized endogenously by liver mainly  
 Rate-limiting step of its synthesis : *HMGCoA reductase* is controlled by :  
     feedback inhibition by cholesterol  
     therapeutic agents - statins  
 2/3rd cholesterol in plasma exists as cholesterol ester  
 Excreted in bile as such or after metabolism to bile acids  
 If excess amount of cholesterol is excreted in bile, it results in gallstone formation.

## Phospholipids

Complex lipids

Similar to TG containing a phosphate group and N-base in place of one FA residue.

## Lipoproteins

Lipoproteins are transport form of lipid and contain lipid in association with proteins. Lipoprotein particle consists of non-polar core of TG and CE surrounded by polar groups of phospholipids, proteins and free cholesterol (Table 7.4, 7.5).

**Table 7.4:** Main groups of lipoproteins

Particle	Property	Electrophoretic mobility	Source
Chylomicron (CM)	Largest Least dense	Origin	Intestine
VLDL	Formed from TG	Pre - $\beta$	Liver
IDL	Intermediate form	broad $\beta$	Catabolism of VLDL
LDL	Principal carrier of cholesterol	$\beta$	Catabolism of VLDL via IDL
HDL	Nascent HDL is disc-shaped which is converted to HDL <sub>2</sub> and HDL <sub>3</sub>	$\alpha$	Liver, intestine

**CHAPTER 8**

# D`Uga U'DfchY]bg

**Learning objectives:**

1. Define different categories of plasma proteins.
2. Discuss functions of all plasma proteins.
3. Explain albumin in detail.
4. Describe proteinuria and its causes.

More than hundred different proteins are present in blood known as *Plasma Proteins*. Examination of proteins in plasma can provide information reflecting disease states in many different organ systems.

Different categories of plasma proteins can be:

- Acute phase reaction proteins
- Carrier proteins
- Fibrinogen
- Coagulation factor
- Complement
- Immunoglobulins
- Enzymes
- Enzyme inhibitors
- Transport proteins.

<i>Functions of proteins</i>	
	<b>Example</b>
Acute phase reaction	Alpha-1 AT Haptoglobin Ceruloplasmin CRP
Buffering	All proteins
Oncofetal	AFP
Complement	C4, C3
Enzymes	Albumin Renin Coagulation factors
Immunity	Complement Immunoglobulin
Coagulation	Clotting factors Fibrinogen

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Enzyme inhibitors	Alpha-1 AT Antiplasmin Antithrombin III
Transport proteins	Albumin Thyroxine binding protein Apoproteins Transferrin Erythropoietin
Hormones*	

\* Other protein hormones circulating in blood are not designated as plasma proteins. Even ferritin is not a plasma protein.

**Table 8.1:** Proteins in other body fluids

Fluid	Condition
CSF	↑ in infection Tumor Hemorrhage
Amniotic fluid	Demyelinating diseases AFP Acetylcholinesterase Fetal Hb
Saliva	Secretory IgA
Feces	α <sub>1</sub> AT
Pleural fluid and ascitic fluid	↑ in infection Malignancy

## Plasma Proteins

Plasma contains water, electrolyte, metabolites, nutrients, proteins and hormones.

Salient Features of Plasma Proteins

1. Proteins in plasma can be simple, conjugated (glycoprotein, lipoproteins).
2. Concentration of protein in plasma determine distribution of fluid between blood and tissues.
3. Plasma proteins exert osmotic pressure.
4. Most plasma proteins are synthesized in liver.
5. Most plasma proteins are glycoproteins, and removal of terminal sialic acid residues from them shorten their half-life (Table 8.2, 8.3).

**Table 8.2:** Total plasma protein concentration

Increase	Decrease
↑ Synthesis: Hypergammaglobinemia Paraproteinemia	↓ Synthesis: Malnutrition Malabsorption Chronic liver disease Immunodeficiency
↓ Blood volume: Hemoconcentration Dehydration	↑ Blood volume: Overhydration Increased permeability: Septicemia Hypoxia
↑ Loss/catabolism: Catabolic state Protein losing states: Enteropathy Nephrotic syndrome Exfoliative disorder of skin, e.g. pemphigus Extensive burns Hemorrhage	

**Table 8.3:** Principal plasma proteins

Prealbumin	
albumin	
$\alpha_1$ -globulin	$\alpha_1$ -antitrypsin
$\alpha_2$ -globulin	Haptoglobin
	$\alpha_2$ -macroglobulin
	Ceruloplasmin
$\beta$ -globulin	Transferrin
	LDL
	C3
$\gamma$ -globulin	IgG, IgA, IgM, IgD, IgE

**Table 8.4:** Other plasma proteins

Plasma proteins	Functions	↑	↓
$\alpha_1$ acid glycoprotein	Glycoprotein group of protein: control bioavailability of drugs downregulation of immune response	Acute phase response by glucocorticoides (GC)	by estrogen nephrotic syndrome
$\alpha_1$ AT	Serpin (serine protease inhibitor) inhibit leukocyte elastase prevent elastic recoil in bronchial tree	Acute phase response estrogens	Genetic deficiency: emphysema Hepatic damage Wilson's disease
Ceruloplasmin	$\alpha_2$ globulin, contains 95% of total serum copper regulates ionic state of iron ( $Fe^{2+} \rightarrow Fe^{3+}$ )	Acute phase estrogen	—
CRP	Activator of classical complement system	Acute phase response	—
Haptoglobin	Binds Hb irreversibly	Acute phase response protein-losing syndrome GC, androgens	genetic deficiency estrogen hemolytic disease hepatocellular disease
$\alpha_2$ macroglobulin	Plasma proteinase inhibitor inhibits kinin, complement, coagulation, fibronolytic pathways	Estrogen nephrotic syndrome	genetic deficiency pancreatitis prostate CA
$\beta_2$ microglobulin	It is light or $\beta$ -chain of HLA on cell surface	Renal failure neoplasm inflammation	—
Retinol binding protein	Binds all trans-retinol prevents retinol from being filtered by glomeruli	Chronic renal disease Diabetic nephropathy	Liver disease protein - malnutrition acquired immunodeficiency

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# Nucleic Acid

## Learning objectives:

1. Explain metabolism of purines and its clinical importance.
2. Describe metabolic disorders of Purine Catabolism.
3. Discuss Diagnostic application of Nucleic Acid.

### *Purine and urate metabolism*

Abnormalities of purine metabolism commonly found in clinical practice, include gout, hyperuricemia, Lesch Nyhan's syndrome, adenosine deaminase deficiency.

## Purine Metabolism

*Humans catabolize purines to uric acid.*

Uric acid is end-product of purine degradation in humans. Since humans lack uricase (that converts uric acid to water-soluble allantoin), their uric acids are higher.

Purine synthesis and breakdown occurs in all tissues, urate is produced only in tissues containing xanthine oxidase, i.e. liver and small intestine (Table 9.1).

**Table 9.1:** Factors affecting urate production

Dietary intake of purine
Rate of purine biosynthesis, degradation and salvage
Renal function
Age, sex
Alcohol intake

Four steps of renal handling of urate (Fig. 9.1)

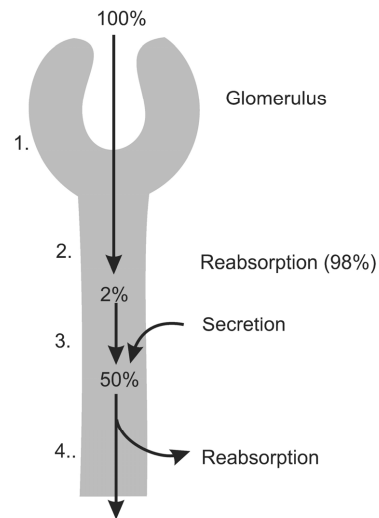
1. Glomerular filtration
2. Reabsorption in PCT (proximal convoluted tubule)
3. Secretion into distal portion of PCT
4. Reabsorption in DCT (distal convoluted tubule)
5. 75 percent of uric acid is excreted in urine and rest is eliminated through intestine.

## Properties of Uric Acid (UA)

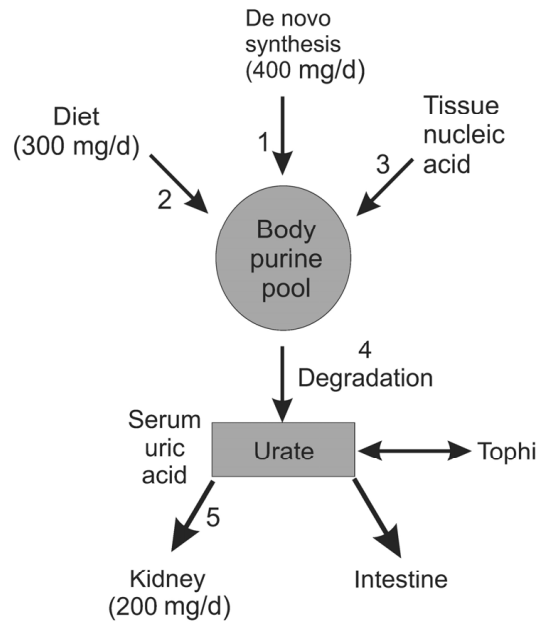
- Weak acid, pKa of 5.75 and 10.3
- Exists predominantly as ionized form: urate



- 98 percent exists as monosodium urate at pH 7.4
- UA is poorly soluble in plasma and urate is ionized
- Ionization of UA decreases as pH falls and becomes less soluble.



**Fig. 9.1:** Renal handling of urate



**Fig. 9.2:** Total body urate pool and causes of hyperuricemia

## Hyperuricemia

Defined as a plasma/serum urate concentration greater than 7 mg/dL.

### Causes of Hyperuricemia

*Causes:* According to underlying physiology:

1. Increased synthesis of purines
2. Increased intake of purines
3. Increased breakdown (turnover)
4. Increased rate of urate formation
5. Decreased rate of excretion (Table 9.2, Fig. 9.2).

**Table 9.2:** Causes of hyperuricemia

Cause	Reason
I. Increased formation of UA	
a. Primary Increase	
i. Purine synthesis	<i>de Novo</i> : ↑ PRPPS*
	Salvage: ↓ HGFRT*
ii. Inherited metabolic disease (IMD)	Insulin/glucagon
GSDI	↑ ATP degradation
	Lactic acidemia block uric acid excretion
	Metabolism of G-6-P through PPP, ↑ R-5-P Purine synthesis
	More G-6-P available diverted to HMP shunt:
	↑ R-5-P Purine
F1P Aldolase deficiency	aldolase deficiency
	(both mechanisms)
b. Secondary increase	
i. Increased dietary intake of purines	Purines oxidized to urate
ii. increased nucleic acid turnover	
Cause trauma surgery Malignancy, starvation, Psoriasis cytotoxic drugs	Purines oxidized to urate
iii. Altered ATP metabolism	
causes:	
Alcohol, tissue hypoxia	Hypoxia and ↓ ATP synthesis, causing lactic acidosis which blocks UA excretion, alcohol ↓ excretion of UA

\* HGPRT deficiency: ↓ activity of salvage pathway causing decreased purine utilization and ↑ uric acid synthesis (Fig 9.3)

\*\* PRPPS: ↑ PRPPS result in increased purine synthesis and increased uric acid synthesis

Contd...

**CHAPTER 10**

# Inherited Metabolic Diseases

**Learning objectives:**

1. Enlist and describe all Inherited Metabolic Disorders with Clinical Laboratory Assessment.
2. Discuss Autosomal Recessive Inherited Metabolic Disorders.
3. Explain Metabolic Disorders caused as a result of mutation.

**Inherited Metabolic Disorders IMD**

Inborn metabolic disorders are caused by genetically determined abnormalities that lead to a block in one of many interrelated biochemical pathways.

The block frequently leads to accumulation of substrate and secondary metabolites or to deficiencies of downstream products (Table10.1).

**Features**

1. IMD are categorized by type of metabolic pathways that are impaired, such as disorders of carbohydrates, fats or amino acids.
2. These disorders may involve pathways of specific cellular organelles, including lysosomes, mitochondria, peroxisomes.

**When to Suspect an IMD?**

When there are unusual, unexplained clinical features or abnormal lab findings in infancy or childhood, the possibility of an IMD should be considered. Also, history of consanguineous marriage or more than one infant in family affected by same abnormality.

*Cause of presentation* can be due to (i) direct or enzyme abnormality and (ii) can be demonstrated indirectly.

**Disorders of Intermediary Metabolism**

It occurs as a result of three basic mechanisms:

1. *Enzyme deficiency* with defective substrate conversion (Table10.2).
2. *Membrane transport defect* resulting in failure of absorption or excessive excretion of a compound (Table10.3).
3. *Defects in receptors* involved in mediating metabolism.

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## Biochemical Basis of Resulting Disease (Fig. 10.1)

1. Accumulation of a substrate to toxic levels ( $\uparrow$ A)
2. Deficiency (due to lack of production) or excessive loss of desired compound ( $\downarrow$ B)
3. Conversion of accumulated compound to altered/toxic metabolite ( $\uparrow$ C)

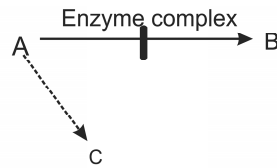


Fig. 10.1: Basis of IMD

Inherited metabolic diseases are the result of gene mutations causing production of an abnormal protein or preventing synthesis of a protein (Table 10.1-10.5).

**Table 10.1:** Inherited metabolic diseases

$\alpha_1$ antitrypsin deficiency Cystic fibrosis Classic galactosemia Familial hypercholesterolemia Glycogen storage disease Hereditary fructose intolerance Lesch-Nyhan syndrome Liver: (disorders of conjugation and excretion of bile:) Gilbert, Crigler-Najjar, Dubin Johnson, Rotor syndrome Hemochromatosis: Lysosomal storage disease Maple syrup urine disease Phenylketonuria Porphyrrias Renal glycosuria Steroid 21-hydroxylase deficiency Tyrosinemia type 1 Vitamin D-dependent and resistant rickets Wilson's disease
---

**Table 10.2:** IMD

	<i>Diseases</i>	<i>Inheritance</i>	<i>Defect</i>	<i>Clinical Features</i>	<i>Diagnostic approach/detection</i>
1.	$\alpha_1$ -antitrypsin	Homozygosity for Z allele	Single amino acid	Liver or lung disease	Phenotyping screening of

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Diseases	Inheritance	Defect	Clinical Features	Diagnostic approach/detection
	(PiZZ)  Heterozygotes (PiMZ)	Substitution	Emphysema and risk increased In smokers Liver: Neonatal Hepatitis Cirrhosis Lung disease	affected individual's  members antenatal screening using fetal DNA from Chorionic Villous
2. Cystic fibrosis	Familial	Impaired Cl <sup>-</sup> Transport Due to CFTR Gene defect (cystic fibrosis transmembrane Conductance receptor) involved in chloride transport, deletion of single codon: common:	affected children develop recurrent respiratory infections irreversible lung disease pancreatic insufficiency leading to malabsorption Intestinal obstruction in neonates infants: failure to thrive vomiting hepatosplenomegaly	α1 AT levels high sweat sodium high plasma immunoreactive trypsin molecular genetic analysis of CFTR gene
3. Classical Galactosemia	Autosomal recessive	Δ F 508 Galactose-1-phosphate Uridyl transference absent (GALT)		plasma galactose ↑ urine galactose positive GALT deficiency in RBCs & chorionic villi
	Adults	cataract	Jaundice cataract, Impaired renal function	
4. Familial Hypercholesterolemia	Autosomal dominant	defect in uptake and catabolism of LDL	cholesterol deposition in skin, tendon, arteries	high serum HDL, total cholesterol
5. Glycogen Storage Disease Type I		Glucose-6-phosphatase deficiency	fasting hypoglycemia accumulation of glycogen in liver and hepatomegaly	liver biopsy for G-6-Pase enzyme deficiency lactacidosis common hyperlipidemia hyperuricemia

Contd...

# Oral Manifestations of Inherited Disorders

## Learning objectives:

1. Describe oral manifestation of all Metabolic Disorders.

## •••Hypophosphatemic Vitamin D Resistant Rickets

X-linked AD.

### Clinical Features

- Hypophosphatemia
- Rickets or osteomalacia
- Resistant to usual doses of vitamin D.

### Causes of Hypophosphatemia

- Defective reabsorption and increased excretion of phosphate by renal tubules
- Rickets secondary to biochemical abnormality.

### Oral Manifestations

Gingival and periapical abscess without concurrent evidence of caries or tooth fracture.

### Systemic Manifestation

- Rickets
- Osteomalacia.

### Laboratory Investigations

- Serum Calcium                   n
- ALP                                   n/↑
- Serum Phosphate.               ↓

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## **Pseudohypoparathyroidism (PHP) and Pseudopseudo-hypoparathyroidism (PPHP)**

PHP                    Hypocalcemia  
PPHP                 Normocalcemia  
X-linked AD

*Causes:* Lack of end-organ receptor for PTH.

### **Oral Manifestation**

- Enamel covered with pits
- Pulp stones near pulp horns
- Retarded root development with large root canals.

### **Systemic Manifestation**

#### *PHP*

- Short stature
- Obesity
- Mental retardation.

#### *PPHP*

- Tall
- Less obese.

### **Common Features**

- Short metacarpals
- Calcification in subcutaneous tissues of scalp and extremities.

### **Laboratory Investigations**

- S. Ca low
- S. PO<sub>4</sub> high.

## **Fabry Syndrome**

X-linked AR  
 $\alpha$ -galactosidase deficiency.

### **Oral Manifestations**

Angiokeratoma of lips (lower lip).

### **Systemic Manifestations**

- Angiokeratoma of skin
- Pain in hand and feet.

### **Clinical Features**

- Pain in extremities
- High blood pressure
- Cardiomegaly
- Albuminuria.

### **Laboratory Findings**

- Proteinuria
- Anemia
- Uremia
- Urine-birefringent lipid inclusion.

### **Lesch-Nyhan Syndrome**

- X-linked AR
- HGPRTase deficiency
- Hyperuricemia
- Choreoathetosis
- Mental retardation
- Self-mutilation.

#### *Facies*

- Distorted due to chewing of lips
- Multiple scars on ears, face, head due to self-inflicted trauma.

#### *Oral Manifestation*

Self-chewing of lips.

#### *Systemic Manifestations*

- Mental retardation
- Amputation due to chewing of distal phalanges.
- Choreoathetosis
- Quadriparesis.

#### *Laboratory Investigations*

Urine: Uric acid raised.

### **MPS VIII**

AR

- $\beta$ -glucuronidase absent
- Accumulation of acid mucopolysaccharides.



# Heme, Iron and Porphyria

## Learning objectives:

1. Explain the synthesis of Porphyrin and describe Porphyria.
2. Discuss Iron in detail.
3. Enumerate all abnormal derivatives of Heme.

## Porphyrin

A cyclic compound with four pyrrole rings connected by methenyl bridges. The arrangement of four nitrogen atoms in the center of porphyrin ring enables porphyrins to chelate various metal ions, e.g.:

Iron chelate : Iron porphyrin—heme  
Magnesium : Magnesium porphyrin—chlorophyll

Metal-free porphyrins have no biological function in humans and protoporphyrin that contains iron is HEME.

## Hemoproteins

*Proteins that contain heme are hemoproteins (Table 12.1).*

Examples:

- Hemoglobin (Hb)
- Myoglobin (Mb)
- Cytochrome C (Cyto C)
- Cyt P<sub>450</sub>
- Catalase
- Tryptophan pyrrolase.

**Table 12.1:** Hemoproteins

<i>Proteins</i>	<i>Function</i>
Hb	Transport of oxygen in blood
Mb	Storage of oxygen in muscle
Cytochromes: Cyt C	Electron transport chain
Cyt P <sub>450</sub>	detoxification of xenobiotics (hydroxylation)

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Enzymes: Catalase	Uses $H_2O_2$ as electron donor and acceptor to form water and oxygen (degrades $H_2O_2$ )
Tryptophan pyrrolase (oxygenase)	Cleaves indole ring with incorporation of oxygen to form N-formylkynurenine (Tryptophan catabolism)

## Heme

Heme is synthesized from porphyrin and iron and is type III porphyrin belonging to series IX. (Fig. 12.1, Table 12.2)

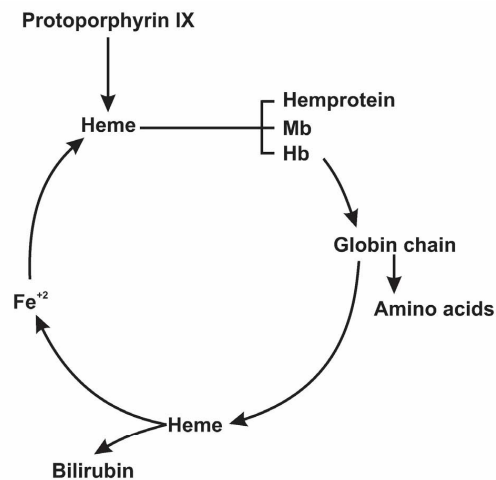


Fig. 12.1: Heme formation and breakdown

Table 12.2: Fact-file—heme

### 1. Synthesis

85% heme synthesized in erythroid cells → Heme for

**Rest:** Heme produced in liver → cyt P<sub>450</sub> enzyme

### 2. Breakdown

Under physiological condition,  $1-2 \times 10^8$  RBCs destroyed/hour

70 kg adult turns ~ 6 g Hb/day

IgHb yields 35 mg bilirubin.

Heme synthesis begins with succinyl CoA and glycine and ends with insertion of Fe<sup>2+</sup> into molecule of protoporphyrin using eight enzymes (four mitochondrial and four cytosolic). Each step produces a progressively less water-soluble molecule until, with protoporphyrin and heme which are insoluble in water.

Thus, excess intermediates early in the pathway tend to appear in urine, while intermediates late in the pathway (particularly protoporphyrin) are found in blood, bile and feces and not in urine.

Porphyria are rare group of disorders of heme synthesis resulting from deficiency of one of the enzymes of this pathway (Fig.12.2).

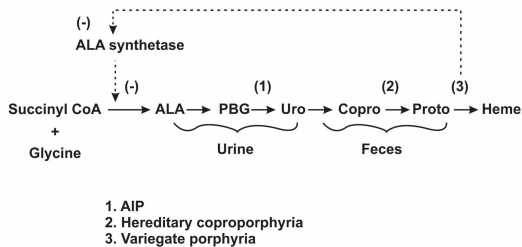


Fig. 12.2: Types of porphyria

## Porphyrin Synthesis (Fig. 12.3)

### ❖ Types of Porphyria

- Inherited
- Acquired.

### ❖ Classification of Porphyria

- Hepatic
- Erythropoietic

### ❖ Types of Porphyria

- Acute
    - Acute intermittent porphyrias (AIP)
    - (HCP) Hereditary coproporphyria
    - (VP) Variegate hepatic porphyria
  - Chronic
    - Cutaneous hepatic porphyria (CHP)
    - Congenital erythropoietic porphyrias (EP)
    - Erythropoietic protoporphyria (EPP)
- } hepatic
- } erythropoietic

# Nutrition

## Learning objectives:

1. Discuss the causes, classification and signs of Malnutrition.
2. Illustrate Nutraceuticals.
3. Describe Nutritional Anemias.
4. Explain Vitamins, functions and deficiency symptoms.

For normal growth and development and maintenance of health, adequate intake of nutrients is essential (Table 13.1)

**Table 13.1:** Essential nutrients

Water	
Carbohydrates	
Proteins (amino acids)	
Fats	Fat soluble : A, D, E, K
Vitamins:	Water soluble : B, C
Minerals:	Na, K, Ca, Mg, P, Cl
Trace elements:	Cu, Fe, Zn, Cr, Mn, Se, I, F, Mo

Malnutrition is suboptimal nutrition arising from inadequate or unbalanced intake, bioavailability of nutrients which leads to biochemical and metabolic consequences (Table 13.2).

In malnutrition, deficiency of energy, protein, vitamin and minerals can occur.

Two forms of protein energy malnutrition are marasmus and kwashiorkor (Table 13.3).

	<i>Marasmus</i>	<i>Kwashiorkor</i>
Cause	Inadequate intake of protein and calories	Inadequate intake of proteins after trauma or infection
Muscle fat	Loss of sub-cutaneous tissue	Muscle and subcutaneous tissue preserved
Weight	Loss of weight	Weight loss, edema
S. albumin	S.albumin normal	S. albumin low (a state of PEM, where adaptation has failed).

In malnutrition, patient shows features of deficiency of energy, protein, vitamins and mineral.

**Malnutrition: Causes**

<i>Causes</i>	<i>Lab investigations</i>
1. Fat malabsorption	Fecal fat Breath Test ( <sup>14</sup> C glycocholate) B <sub>12</sub> absorption test Pancreatic enzymes in stool : trypsin, chymotrypsin, elastase microscopic examination for fat droplets
2. Protein malabsorption	Serum protein α - 1 anti-trypsin in feces α -anti-trypsin clearance by intestine
3. Celiac disease	Visualization technique antigliadin (AGA) Ab assay in serum anti- <i>endomysium</i> (AEA) Ab in serum antireticulin (ARA) Ab in serum
4. Carbohydrate malabsorption	D-Xylose absorption test Breath H <sub>2</sub> after lactose administration
5. Vitamin malabsorption	B <sub>12</sub> malabsorption, correction by pancreatic enzyme administration Lab investigation of megaloblastic anemia Vitamin A and D absorption test
5. Mineral malabsorption	<sup>59</sup> Fe absorption

**Table 13.2:** Classification: Malnutrition

<b><i>I. On the basis of lesion</i></b>	
1. Gastric:	Deficiency of intrinsic factor causing defective absorption by vitamin B <sub>12</sub> .
2. Small intestine: Luminal	
a. Liver	Cirrhosis of liver Extrahepatic biliary obstruction causing defective bile secretion
b. Pancreas	Pancreatitis CA pancreas Zollinger-Ellison Syndrome
c. Brushborder defect	
i. Lesion of mucosa:	Celiac disease Gluten-induced enteropathy Tropical sprue Infiltration by tumor Sarcoidosis Radiation Drug-induced Infection Parasite

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ii. Inborn errors:	Amino acid transport defect Glucose transport defect Abetalipoproteinemia
<b>II. On the basis of nutrient</b>	
1. Fat malabsorption	Steatorrhea Hepatobiliary disease Pancreatic disorder Brushborder lesion
2. Carbohydrate malabsorption	Brushborder lesions Enzymatic defect: Disaccharidase deficiency (maltase, sucrase, lactase)
3. Protein malabsorption	<i>Protein losing enteropathy:</i> Tropical sprue Colitis Crohn's disease Infantile allergy Amino acid transporter defect
4. Malabsorption of vitamin	Water-soluble vitamin: B (pernicious anemia: IF deficiency) Fat-soluble vitamin: In fat malabsorption
5. Malabsorption of minerals	
Iron:	Parasite infestation Phytates, oxalates

**Table 13.3:** Signs associated with malnutrition

Organ	Signs	Deficiency
Face	Diffuse depigmentation	Riboflavin
Skin	Purpura	Vitamin K
	Hyperpigmentation	Cobalamin, folic acid
	Hyperkeratosis	Retinol, EFA
		Ascorbic acid, niacin
Eyes	Petechial hemorrhage	
Lips	Poor dark adaptation, Bitot's spots	
	Angular stomatitis, cheilosis	
Tongue	Raw tongue	Retinol
	Scarlet tongue	Riboflavin or niacin
	Glossitis	Riboflavin
		Riboflavin
	Atrophic filiform papillae	Niacin, riboflavin, folic acid, cobalamin, iron
Teeth	Mottled enamel, caries	Niacin, folate, riboflavin, iron, cobalamin
Gums	Pale	Fluoride
Nails	Koilonychia, pallor	Ascorbic acid
CVS	Cardiomegaly, wet beri-beri	Niacin
Nervous system	Neuropathy, sensory and motor loss,	Thiamin
	dry beri-beri	Thiamin

# Malignancy

## Learning objectives:

1. Describe Endocrine Syndromes.
2. Illustrate Biochemical Changes in Cancer.
3. Discuss MEN and differentiate between MEN I and MEN IIa.

## Malignancy

Term *paraneoplastic syndrome* includes all the systemic manifestations of cancer not directly related to presence of primary tumor, whether or not they are due to a hormone termed as *Ectopic hormone* (Table 14.1).

Most of these endocrine syndromes occur with tumors of non-endocrine tissue which have embryonic origin from neural crest cells. However, non-APUD tumors do occur.

**Table 14.1:** Endocrine syndromes

	<i>Syndrome</i>	<i>Tumor</i>	<i>Hormone</i>
1.	Cushing's syndrome	Small cell CA bronchus Carcinoid tumors	ACTH
2.	Hypercalcemia of malignancy*	Non small cell CA lung Renal adeno CA	PTHrP
3.	Hyponatremia (SIADH)	Small cell CA lung	Vasopressin
4.	Hypoglycemia	Mesenchymal tumors (non $\beta$ cell tumors) Small cell cancer	Insulin-like growth factors (IGFs)
5.	Gynecomastia	Lung testicular cancer Hepatic tumor Carcinoid	hCG
6.	Ectopic acromegaly	Small cell lung CA	GHRH GH
7.	Polycythemia	Pancreatic islet tumors Uterine fibromyomata Cerebellar hemangioblastoma	IGF - I Erythropoietin

\* Hypercalcemia is common in hematological malignancies, particularly myeloma due to release of osteoclast-activating cytokines (IL-1, TNF $\beta$ ) by tumors.

## Metabolic Effects of Tumors

Neoplastic cells differentiated tissues sometimes synthesize compounds that are not normally produced from that tissue. These substances may alter the metabolism and may produce clinical effects and hormone syndromes. Also, these substances may be biologically inactive and serve as *tumor markers*.

Also, neoplasias are associated with rare syndromes with endocrine or neurotransmitter properties showing common tumors, cytological characteristics, e.g. in *APUDomas* staining technique show ability for *Amine Precursor Uptake and Decarboxylation* (Table 14.2, 14.3).

Certain tumors secrete physiologically active amines, while others secrete peptide hormones (e.g. pituitary, parathyroid and C-cells of thyroid).

## Biochemical Changes in Cancer

Metabolic changes in cancer are not always due to aberrant hormone secretion, but may be due to some other effect of tumor or as a consequence of treatment.

1. *Serum:* *High:* Calcium, uric acid, lactate, globulin, potassium, phosphate  
*Low:* Fasting glucose, albumin, magnesium
2. *Anemia:* Common, could be due to hemorrhage, hemolysis, malnutrition or bone marrow infiltration by tumor cells.
3. *Hypercoagulable* state resulting in thromboembolism.
4. *Cancer cachexia:* A syndrome of weakness and generalized wasting. Causes can be deficient food intake, intestinal obstruction or anorexia of malignancy, malabsorption, increased demands and consumption of nutrients by tumor, release of cachectin (TNF $\alpha$ ). The condition is rarely reversible unless the underlying tumor can be treated successfully.

**Table 14.2:** Carcinoid tumor

Site	:	Enterochromaffin cells of gut: cells of APUD series 90% in appendix and ileocecal region Others: Gut, gallbladder, pancreatic duct, bronchi
Property	:	Low grade malignancy Locally invasive Rarely metastasize
Clinical features	:	GIT: nausea, vomiting, diarrhea, colic Cardiovascular: flushing, hypotension Respiratory: bronchospasm Pellagra
Others	:	Manifestation of secretion of other hormones Vasoactive amines: serotonin mainly Histamine
Mediators	:	Kinin Hormones ACTH 5 HIAA in urine Plasma 5 HTP, histamine, peptide hormone
Lab. diagnosis	:	



**Table 14.3:** Origin of APUD tumors

<i>Origin</i>	<i>Tumor</i>
GIT	Carcinoid
Pancreas	Islet cell CA
CNS	Ganglioneuroblastoma
	Neuroblastoma
Thyroid	Medullary thyroidal CA
Skin	Melanoma
Adrenal medulla	Pheochromocytoma
Lung	Small cell CA
	Carcinoid tumor

**Argentaffin cells:**  
 Round and stain with silver salts  
 Derived from embryonic gut  
 Present in ileum, appendix in abundance  
 so, present in pancreas, stomach, rectum

## Multiple Endocrine Neoplasia (MEN)

MEN are familial disorders with autosomal dominant inheritance, where tumor hyperplasia occurs in two or more endocrine glands (Table 14.4).

<i>MEN I</i>	<i>MEN IIa</i>
Parathyroid	Thyroid (medullary cell CA)
Pancreas	Adrenal medulla (pheochromocytoma)
Pituitary (anterior)	Parathyroid CA
Thyroid (follicular cell)	<b>MEN II b</b>
Adrenal cortex	Thyroid (medullary cell CA)
	Adrenal medulla (pheochromocytoma)
	Mucosal neuroma
	may be associated with Marfanoid, Hirschsprung disease

**Table 14.4:** Pancreatic tumors

<i>Pancreatic islet tumors</i>	<i>Hormone produced</i>
Zollinger-Ellison syndrome	Gastrin
Peptic ulcer gastrinoma	Insulin
Insulinoma - Hypoglycemia	Glucagon
Glucagonoma - Hyperglycemia	VIP
VIPoma - Watery diarrhea, hypokalemia	Somatostatin
Somatostatinoma - diarrhea, steatorrhea, gallstones	Serotonin, PG
DM	Serotonin
Carcinoid syndrome - diarrhea, hepatomegaly	Histamine
	Kinin
	ACTH

**Questions:**

- 1) What are the metabolic effects of tumors?
- 2) Discuss the biochemical changes in cancer?
- 3) What is Carcinoid tumor?
- 4) Explain Multiple Endocrine Neoplasia (MEN) in detail.
- 5) Name some endocrine syndromes and hormones associated with it.

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# Toxicology

**Learning objectives:**

1. Enlist Toxic agents and discuss its features.
2. Describe Forensic Toxicology
3. Illustrate common methods for analysis of different drugs.
4. Explain Toxicology disciplines.

Toxicology includes analysis of drugs, heavy metals and other chemical agents in body fluids and tissues for three main purposes:

1. Diagnosis and management in drug-overdose (Table 15.1, 15.6)
2. Therapeutic monitoring of drug (TDM) (Table 15.2-15.4)
3. Screening for the presence of drugs of abuse

**Table 15.1:** Reasons for high serum drug concentrations

Patient noncompliance
Improper dosage
Poor bioavailability of preparation
Malabsorption
Drug interaction
Liver disease
Kidney disease
Altered protein binding
Genetic factors affecting metabolism

**Table 15.2:** Need for monitoring of drug therapy

- |   |
|---|
| <ol style="list-style-type: none"><li>1. Patient taking more than one drug has increased probability of drug interaction and toxicity</li><li>2. To follow patient compliance</li><li>3. Indicated in drug groups which show individual variation or are toxic above therapeutic range like:<ul style="list-style-type: none"><li>Antiepileptics</li><li>Theophylline</li><li>Tricyclic antidepressants</li><li>Lithium carbonate</li><li>Aminoglycosides</li></ul></li></ol> |
|---|

**Table 15.3:** Drugs requiring TDM

	<i>Drug</i>	<i>Therapeutic range (toxic level)</i>
1.	Carbamazapine	4–12 µg/ml (>15.0 µg/ml)
2.	Digoxin	0.5–2 ng/ml (> 2.5 mg/ml)
3.	Cyclosporine	12 hrs after dose 100–400 ng/ml (renal transplant) 24 hrs after dose 100–200 ng/ml (> 400 ng/ml)
4.	Phenytoin	10–20 µg/ml ( > 20 µg/ml)
5.	Phenobarbitone	15–40 µg/ml (> 35 µg/ml)
6.	Theophylline	8–20 µg/ml (>20 µg/ml)

**Table 15.4:** Features of toxic agents

<i>Agent</i>	<i>Properties</i>	<i>Clinical features</i>	<i>Analysis/Measurement</i>
1.	<b>Alcohol</b>		
a.	Ethyl alcohol  Principal pharmacologic action : CNS depression  <i>Legal intoxication :</i> 80–100 mg/dl >400 mg/dl : death  * Can detect alcohol ingestion within previous 8 hours	disorientation or mild euphoria stupor coma	blood alcohol breath alcohol saliva alcohol urine alcohol*
b.	Methyl alcohol  Used as solvent in a number of commercial product  consumed as illicit liquor	severe metabolic acidosis peripheral circulatory failure blindness causes optic neuropathy and blindness death	serum formate
c.	Isopropanol  Used as rubbing alcohol has twice the CNS depressant action as ethanol not as toxic as methanol	similar to ethanol	Urine isopropanol levels Urine acetone
2.	<b>Analgesics</b>		
a.	Acetaminophen (paracetamol)  used in treatment of mild pain or fever	nausea, vomiting, abdominal discomfort <i>over dosage</i> causes hepatic, nephrotoxicity liver toxicity (which can be predicted after 4 hrs post ingestion)	<i>Reference interval :</i> Serum 10–30 µg/ml <i>Toxic level :</i> >200 µg/ml  Immunoassay (EMIT) for acetaminophen
b.	Acetyl salicylic Acid (Aspirin)  Analgesic Antipyretic	<i>Therapeutic</i> <100 µg/ml <i>Toxic</i> >100 GI intolerance	S.salicylate U.salicylate

Contd...

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3.	<b>Barbiturates</b>	Anti-inflammatory  Low therapeutic index  High abuse potential Suppress CNS and neuronal activity sedative and hypnotic	impaired hemostasis, deafness, headache, vertigo, tinnitus, nausea, vomiting, hyperventilation  Coma hypothermia  hypotension cardiorespiratory arrest	(photometric GC HPLC)  S.barbiturate level  (immunoassay)
4.	<b>Benzodiazepines</b>	CNS depressants anxiolytic sedative hypnotic muscle relaxant anticonvulsant "date-rape" pill	drowsiness slurred speech  ataxia coma	S. benzodiazepine U. benzodiazepine (HPLC immunoassay)
5.	<b>Cannabinoids:</b>	<b>Principal cannabinoid is <math>\Delta^9</math>-</b> illegal drug <i>psychoactive effects :</i> euphoria Inhaled in smoke t $\frac{1}{2}$ THC - day in casual user and 3-5 d. in chronic users	drug of abuse in infrequent users : Impairment of intell- ectual performance	THC +ve in urine for 2-5 d after use immunoassay cut off value of 20 ng/ml THC
6.	<b>Cocaine</b>	CNS stimulant Euphoria	mydriasis tachycardia hypertension hyperthermia agitation seizure coma	Urinary benzoylecgonine and ecgonine methyl ester  (Immunoassay GC, HPLC)
7.	<b>Lysergic acid diethylamide (LSD)</b>	Hallucinogenic	mydriasis tachycardia salivation lacrimation vomiting panic attack	Urine LSD immunoassay
8.	<b>Amphetamines</b>	Used to treat obesity CNS stimulant	<b>Overdose :</b> dizziness  tremor hypertension cardiac arrhythmia convulsion coma	Urine Immunoassay GC/MS
9.	<b>Opioids/Opiates</b>	analgesic sedative	sedation euphoria	Immunoassay Urine opiate

Contd...

## **Section III**

# **Endocrines**

# Endocrine Hypothalamus

## Learning objectives:

1. Examine the role of Hypothalamus in body

The brain is also one of the most important endocrinal glands.

## Hypothalamus

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Hypothalamus lies at the base of the brain and contains specialized nerve cells (nuclei) that synthesize hormones to regulate hormone secretion from the pituitary. These hormones are transported along the axon of the nerve terminal to be finally released into portal blood system.

Hypothalamus is composed of three regions:

- Rostral or supraoptic hypothalamus
- Middle or tuberal hypothalamus
- Caudal or mammary hypothalamus

## Biorhythm

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In most endocrine systems, rhythmic secretion of hormones occurs which may vary over minutes to hours, days, weeks, or even longer periods.

### Circadian or Diurnal Rhythm

When rhythm occurs with a time interval of approximately 24 hrs, they are termed as *circadian* or *diurnal*. They are usually regulated by one or more internal “clocks” synchronized by factors such as dark/light cycle and sleep.

### Ultradian Rhythm

They are the biorhythms that occur more frequently than once a day.

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### ❖ Infradian Rhythm

They occur over time periods longer than 24 hours, e.g. menstrual cycle.

Endocrine rhythms involve pulsatile and circadian release of pituitary hormones, e.g. LH (every 90 min) ACTH (peak at 8 am).

#### Questions:

1. Write the functions of Hypothalamus.
2. What is Biorhythm? Write its types.

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# Pituitary Gland

## Learning objectives:

1. Describe Hypothalamic Pituitary axis.
2. Enlist and explain all Pituitary hormones.
3. Discuss all disorders related to Pituitary gland.

## ❖ Pituitary Gland

Pituitary consists of three lobes—anterior lobe, or adenohypophysis (primary glandular tissue), posterior lobe, or neurohypophysis and intermediate lobe. Anterior pituitary grows from primary oral cavity (Rathke's pouch) and consists of glandular tissue; posterior pituitary is neural in origin (Table 17.1).

**Table 17.1:** Pituitary hormones

<i>Adenohypophysis :</i>	GH Prolactin (PRL) Thyroid stimulating hormone (TSH) Follicle stimulating hormone (FSH) Luteinizing hormone (LH) Adrenocorticotrophic hormone (ACTH) and $\beta$ -Lipotropin (LPH)
<i>Neurohypophysis hormones:</i>	ADH (Antidiuretic Hormone) Oxytocin

*Flow sheet of trophic hormones and target gland hormones*

Hypothalamic hormone	GnRH	TRH	SS (GHRH)	GHRH	CRH	AVP ADH
Pituitary hormone	FSH, LH	TSH	-	GH	ACTH	-
Target gland hormone	Estrogen Progesterone Testosterone	T3, T4	-	-	Corticosteroids	

## ❖ Combined Test (Triple Bolus Test) of Anterior Pituitary Function

### Procedure

In a overnight fasting patient, take basal blood sample.

Analyze for glucose, cortisol (or ACTH), FSH, LH, TSH, fT4, GH.

Give 200  $\mu$ g TRH, 100  $\mu$ g GnRH, 1.5 U/kg body weight soluble insulin.

Take blood samples at various interval and Table 17.2 describes assessment of pituitary function.

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Analysis					
Time (min)	Glucose	Cortisol	TSH	GH	FSH, LH
0	√	√	√	√	√
15	√	-	-	-	-
20	-	-	√	-	√
30	√	√	-	√	-
45	√	-	-	-	-
60	√	√	√	√	√
90	√	√	-	√	-
120	√	√	-	√	-
Response	<2.2	> 200 mmol/L	> 5 mU/L	>20 nmol/L	> 1.5 times, >5 times

Table 17.2: Assessment of pituitary function

<b>Indications</b>	
1. Detection of hormonal deficiency	
a. before and after treatment of pituitary tumors	
b. as a result of pituitary surgery or irradiation.	
2. Recognition of hormone producing tumors	
<b>Methods</b>	
Gland/Organ	Test
a. Before surgery	
Adrenal	Serum cortisol Cosyntropin stimulation test Estradiol in females
Gonads	Testosterone in males
Thyroid	TSH
b. After surgery	
Adrenal	Cosyntropin stimulation test
Gonads	Sex hormones
Thyroid	TSH

## Hypothalamic Pituitary Axis

Hypothalamus synthesizes hormones to regulate hormone secretion from the pituitary (Table 17.3-17.6)

Table 17.3: Test for hypothalamic target gland/organ axis

<b>Hypothalamic-Pituitary-Adrenal axis</b>
Normal morning cortisol
Cosyntropin stimulation test
<b>Hypothalamic-Pituitary-Thyroid axis</b>
Serum free thyroxine
TSH
<b>Hypothalamic-Pituitary-Gonadal axis</b>
Serum gonadotropin (LH, FSH)
Sex steroids (estradiol/testosterone)

**Table 17.4:** Problems in evaluation of hypothalamic pituitary axis

<i>Condition</i>	<i>Reason</i>
Obesity	GH dynamics impaired insulin induced hypoglycemia
DM (Diabetes Mellitus)	GHRH, arginine, levodopa test fail to provoke GH secretion Diabetics have normal or elevated GH levels which do not further rise in response to hypoglycemia, arginine
CRF (Chronic Renal Failure)	Due to prolongation of their half-life, GH, PRL, LH, FSH, TSH and free cortisol levels are elevated
Starvation	Dexamethasone suppression of cortisol impaired With fasting and malnutrition, GH secretion increases
Depression	PRL and TSH dynamics are normal Dexamethasone suppression altered

**Table 17.5:** Endocrine tests of hypothalamic pituitary function

Cosyntropin test (ACTH stimulation) Insulin hypoglycemia test (IHT) Metyrapone test GHRH - arginine infusion test Glucose GH suppression test TRH test GnRH test Clomiphene test CRH test
--

**Table 17.6:** Causes of hypothalamic and pituitary dysfunction

Invasion	Craniopharyngioma Primary central nervous system tumor Epidermoid and dermoid tumors Pituitary adenoma
Infarction	Postpartum hemorrhage Spontaneous hemorrhage
Infiltration	Sarcoidosis Hemochromatosis
Injury	Histiocytosis X Head trauma

Contd...

# Adrenals

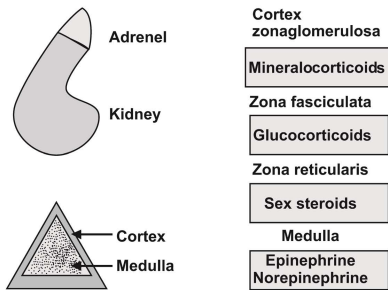


Fig. 18.1: Adrenal cortex and medulla

**Learning objectives:**

1. Explain Pituitary – Adrenal axis.
2. Describe all hormones secreting from Adrenals.
3. Illustrate Cushing’s Syndrome.
4. Discuss all endocrinal disorders associated with Adrenal gland.

Adrenals lie just above kidneys and can be divided into cortex and medulla on the basis of anatomical and functional grounds.  
 Cortex secretes glucocorticoids (GC) mineralocorticoids (MC) and sex steroid hormones which are synthesized from cholesterol obtained from both HDL and LDL in plasma.  
 Medulla secretes, catecholamines, mainly epinephrine.

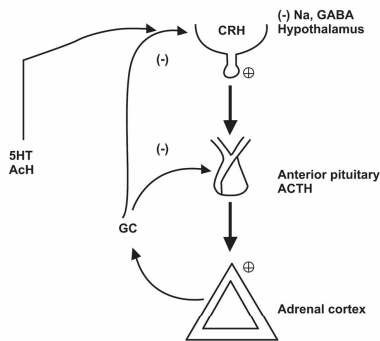


Fig. 18.2: Regulation

Corticosterol binding globulin or transcortin CBG  
 90% CBG bound  
 10% free  
 CBG ↑ pregnancy  
           estrogen treatment  
           ↓ hypoproteinemia  
           (Nephrotic syndrome)

## Adrenal Cortex Hormones (Fig 18.1, 18.2)

Glucocorticoids  
Mineralocorticoids  
Sex hormones

## Adrenal Medulla Hormones

Epinephrine  
Norepinephrine

## Cortisol: Characteristics

Show diurnal variation

Circadian rhythm with peak secretion in early morning and declining with lowest level at midnight.

*Sampling* at 8 to 9 am and 11 pm to detect loss of diurnal variation.

Stresses such as fever, trauma, exercise, anxiety cause increased secretion of CRH (Table 18.3-18.6).

*Caution:* Synthetic glucocorticoid prednisolone cross-reacts with cortisol in immunoassay.

### *Diurnal Rhythm of GC = Biological Clock.*

Biological clock resides in suprachiasmatic area of brain synchronized by exogenous influences of light.

Related to sleeping waking cycle.

Levels are highest at start of working day and 8 am sample cortisol is normal but, 10 pm cortisol falling to lowest with onset of sleep.

**Table 18.1:** Diagnostic test for causes of Cushing's syndrome

	<i>Pituitary macroadenoma</i>	<i>Pituitary microadenoma</i>	<i>Ectopic ACTH or CRH</i>	<i>Adrenal tumor</i>
ACTH	↑ ↑ ↑	n - ↑	↑/↑↑↑	↓
Response to high dose dexamethasone	< 10	95	<10	<10
Response CRH to	> 90	> 90	< 10	< 10

**Table 18.2:** Tests of pituitary-adrenal axis

ACTH stimulation test
IHT
CRH test
Metyrapone test

**Table 18.3:** Adrenal hypofunction

<b>Causes</b>	Glucocorticoid treatment
	Autoimmune
	Infection (TB, viral)
<b>Tests</b>	
	Plasma cortisol : Low
	Synacthen test : Response absent or blunt

**Table 18.4:** Adrenal hyperfunction

<b>Causes:</b>	Cushing's Syndrome Corticosteroid or ACTH treatment Pituitary hypersecretion of ACTH (Cushing's disease)	ACTH
<b>Tests:</b>	Adrenal adenoma Adrenal carcinoma Ectopic ACTH secreting tumors carcinoid, carcinoma bronchus 24 hrs U. Cortisol excretion increased Dexamethasone suppression test: failure to suppress ACTH release IHT: glucose intolerance Loss of diurnal variation of cortisol secretion	

**Cushing's syndrome** (Fig 18.4-18.6)

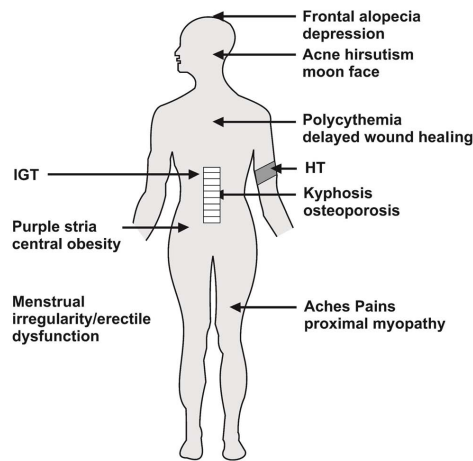


Fig. 18.4: Cushing's Syndrome

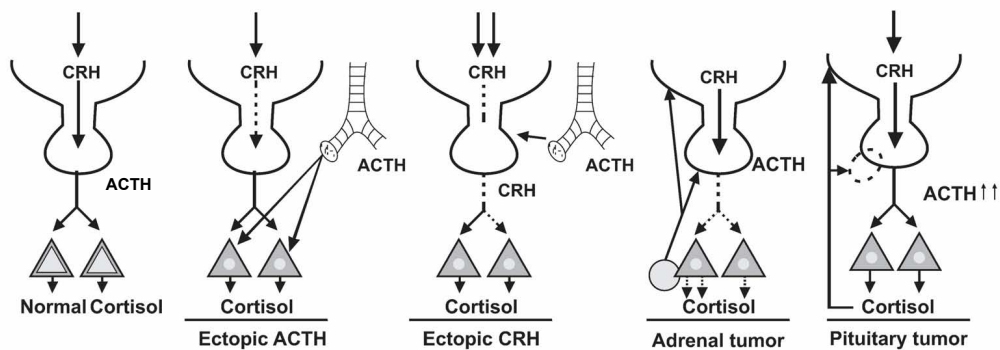


Fig. 18.5: Causes of Cushing's syndrome

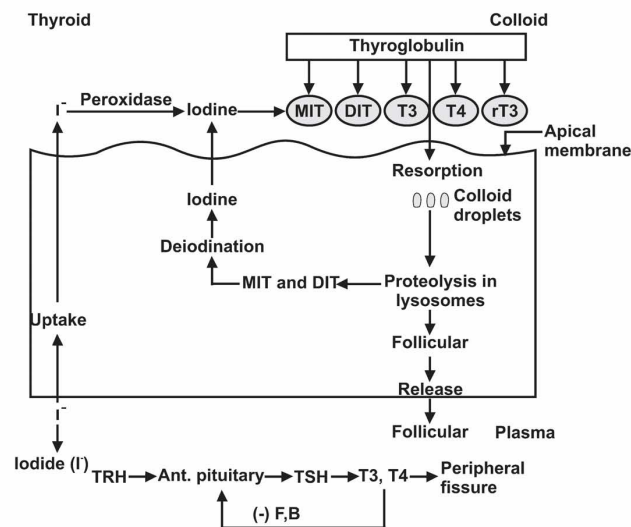
# Thyroid

## Learning objectives:

1. Explain about synthesis of Thyroid hormones.
2. Discuss utility of measuring free T4 and T3.

Hormones secreted	Cells
Thyroxine (T4)	Follicular cell
Tri-iodothyronine (T3)	Follicular cell
Calcitonin	C-cells

While T4 is 10 times more in concentration, than T3, T3 is 3-4 times more potent than T4.



**Fig. 19.1:** Synthesis of thyroid hormones

- **Biologically active hormone T3,**
- 85% T3 in plasma formed by deiodination of T4
- rT3 arise from monodeiodination of T4 and is biologically inactive
- 70% of plasma T4 bound to TBG
- 80% of plasma T3 bound to TBG
- Rest bound to transthyretin and albumin
- Only 0.05% T4 are free
- 0.2% T3 are free

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**Lab investigations**

p TSH  
p T3, T4  
Thyroid antibody  
Serum thyroglobulin  
Thyroid enzyme

**USE OF THYROID TESTS**

- TT3 is reliable measure of hyperthyroidism, but is no value in hypothyroidism.
- Normal TSH rules out primary thyroid dysfunction

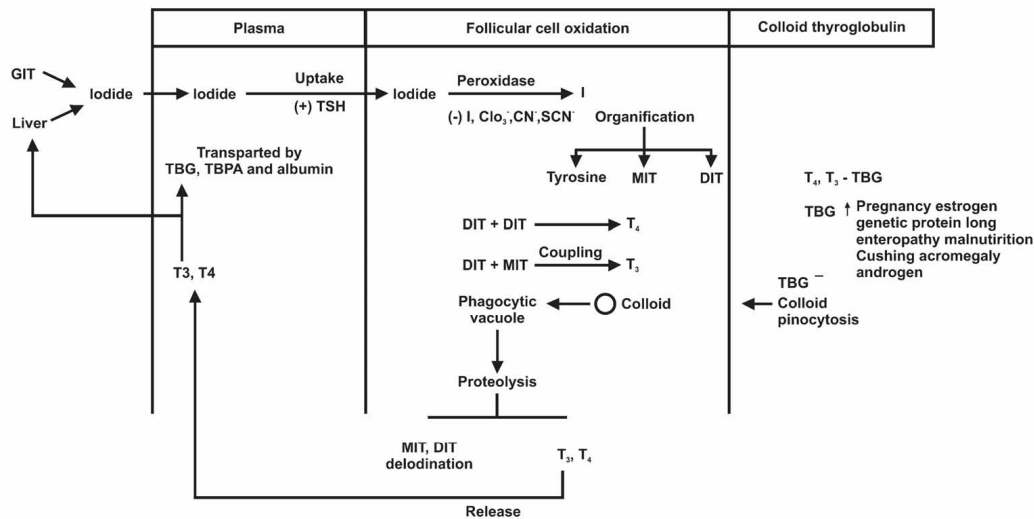
**Utility of measuring free T4 & T3 (fT4, fT3):**

Free T3, T4 concentration are independent of changes in concentration and affinity of thyroid hormones finding proteins and provide reliable measure to diagnose thyroid disorders.

Condition	Reason
Assay interference	Endogenous heterophilic Ab Observed in thyrotoxicosis
Pregnancy (TBG)	1st trimester – weak thyrotropic effect of placental hCG (present in high concentrations) ↑[TBG] during pregnancy results in ↑plasma (TT4) and (TT3) Transient postpartum thyroid dysfunction.
Secondary thyroid disorders:	
Central hypothyroidism	[TSH] normal [FT3], [FT4] low TSH bioactivity reduced
Thyroid tumor (TSH secreting tumor)	↑ [FT3], [FT4] hyperthyroidism ↑ [TSH]
Endorgan resistance	↑ P T <sub>3</sub> , T <sub>4</sub>
Non-thyroidal illness (NTI)	n or ↑ [TSH] Sick-Euthyroid syndrome low [T <sub>3</sub> ] • Alteration in hypothalamus – pituitary thyroid axis (-) (-) TSH release by ↑ dopamine, cytokine, cortisol, somatostatin • Change in TBG and affinity characteristics. • Impaired uptake of thyroid hormone in peripheral tissues ↓ Production of T3 in peripheral tissues. Change in T3 ®
In NIT, TSH < 0.1 m U/L and increase due to recovery	
Drug intake	
G.C. dopamine	
Li, amiodarone, iodide	↓ TSH ↓ Thyroid hormone secretion
Estrogen, tamoxifen, raloxifene, clofibrate	Induce hyperthyroidism
Androgen, GC, anabolic steroids	↓ TBG
Phenytoin, carbamazepine, barbiturates	↑ hepatic metabolism



### ❖ Synthesis



Disorder	Causes	Tests
1. Hyperthyroidism	<ul style="list-style-type: none"> <li>Grave's disease</li> <li>Toxic multinodular goiter</li> <li>Exogenous iodine and iodine containing drugs</li> </ul>	High fT <sub>3</sub> and fT <sub>4</sub> low TSH
2. Hypothyroidism	<ul style="list-style-type: none"> <li>Ectopic thyroid tissue</li> <li>Atrophic hypothyroidism (Hashimoto's) autoimmune</li> <li>Postsurgery</li> <li>Congenital</li> <li>Iodine deficiency</li> <li>Secondary (pituitary hypothalamic)</li> </ul>	high TSH, fT <sub>4</sub> low

## Screening Tests for Thyroid Disease

### Congenital hypothyroidism

- TSH in capillary blood
- T<sub>4</sub>

### Questions:

- Write the mechanism of synthesis of thyroid hormones.
- Describe hyper and hypothyroidism.
- What are screening tests for thyroid disease?

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Section IV

**Pathophysiology**

# Cardiac Function

## Learning objectives:

1. Discuss Cardiac biomarkers.
2. Explain markers in Cardiac failure.
3. Enumerate Cardiovascular risk factors.

Cardiac Marker is a clinical laboratory test useful in detection of acute myocardial infarction (AMI) or minor myocardial injury.

## Cardiac Biomarkers (Table 20.1, 20.2)

Creatine kinase (CK)  
Lactate dehydrogenase (LDH)  
Myoglobin  
Troponin

**Table 20.1:** Cardiac biomarkers

	<i>CK</i>	<i>CK-2</i>	<i>LDH-1</i>	<i>Myoglobin</i>	<i>Troponin</i>
1. Hours until marker increases above upper reference limit	3–8	3–8	8–12	1–3	3–8
2. Hours until peak concentration	10–24	10–24	72–144	6–9	24–48(1st peak) 72–100 (2nd peak)
3. Days until return to within reference interval	3–4	2–3	8–14	1	4–10

**Table 20.2:** Prediction of reperfusion status

<i>Test</i>	<i>Utility</i>
CK-2	Rate of appearance of CK-2 in association with reperfusion occurs 90 minutes after thrombolysis
Troponin Myoglobin	Useful in detecting reinfarction in postinfarct angina

Contd...

Contd..

Myoglobin to total CK ratio > 5 implies reperfusion  
 (specificity of > 96%)  
 (sensitivity of 75%)  
 best indicator for prediction

### Ischemia-Modified Albumin (IMA)

It is an early and sensitive marker of cardiac ischemia. N-terminal of albumin gets altered in myocardial infarction and it no longer can bind to metals such as Cu, Co, Ni. This is termed as *ischemia-modified albumin (IMA)* and can be detected by reduction in albumin's ability to bind cobalt. IMA is promising marker for early detection and has greater sensitivity than ECG and troponin. However, its precise role in patient management remains to be established.

Details of apolipoproteins and lipid profile can be referred in concerned section.

*Marker of deep vein thrombosis: D-dimers*

*D-dimers* are released during conversion of fibrinogen to fibrin (during thrombus formation). Presence of raised values of D-dimer in plasma is diagnostic of presence of a thrombus.

### Cardiovascular risk factors (Table 20.3)

**Table 20.3:** Cardiovascular risk factors

Classic risk factors:  
 i. Abnormal levels of lipids:  
 ii. Cholesterol  
 Triglyceride  
 HDL-C (low)  
 Others:  
 Hypertension  
 Smoking  
 Diabetes mellitus  
 Obesity  
 Family history  
 Recent Biochemical risk factors:  
 Homocysteine  
 C-reactive protein (hs-CRP)  
 Lipoprotein (a)  
 Fibrinogen  
 IMA

### Markers of Cardiac Failure

*Cardiac failure* is a clinical condition associated with signs and symptoms of left or right ventricular dysfunction.

**Natriuretic Peptides** (Table 20.4)

Three natriuretic peptides are:

1. Atrial natriuretic peptide (ANP)
2. Brain natriuretic peptide (BNP)
3. C-type natriuretic peptide (CNP)

**Table 20.4:** Natriuretic peptides

	<i>ANP</i>	<i>BNP</i>	<i>CNP</i>
Site of production	atria	brain cardiac ventricles	brain kidney endothelial cells
Binding to receptors	A-type C-type	A-type C-type	B-type C-type
Function	natriuresis reduce ecf volume	natriuresis reduce ecf volume	vasodilator local tissue action
Utility	diagnostic marker of cardiac failure	prognostic marker in cardiac failure marker of both right and left ventricular dysfunctions	no proven value as a cardiac hormone

<i>Analyte</i>	<i>Assay</i>	<i>Reference range</i>
CRP	Immunoassay	6.8-820 µg/dl
Homocysteine	GC-MS, HPLC	4-12 µmol/L
Fibrinogen	Colorimetric	200-400 mg/dL

**Tests useful in diagnosis and management of hypertension** (Table 20.5)**Table 20.5:** Clinical biochemistry tests useful in diagnosis and management of hypertension

<i>Analyte</i>	<i>To rule out</i>
Glucose	Diabetes mellitus
Urea	Kidney disease
Creatinine	Kidney disease
Urine protein	Kidney disease
Potassium	Primary hyperaldosteronism monitoring treatment
Aldosterone:renin ratio	Primary hyperaldosteronism monitoring treatment
Calcium	Primary hyperparathyroidism
Phosphate	Primary hyperparathyroidism
PTH	Primary hyperparathyroidism
IGF-1	GH excess: acromegaly
OGT	GH excess: acromegaly
Catecholamine metabolites (HMA)	Pheochromocytoma
Urine cortisol	Cushing's syndrome

**Questions:**

1. What are cardiac biomarkers?
2. Define Ischemia Modified Albumin (IMA).
3. Discuss Natriuretic Peptides.
4. Write the name of tests useful in diagnosis and management of hypertension.

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# Liver Functions

## Learning objectives:

1. Illustrate different functions of liver.
2. Discuss Bilirubin metabolism and laboratory tests of Bilirubin.
3. Describe Jaundice and Cirrhosis in detail.
4. Explain different tests done in liver disease.

## Liver

### Functions of Liver

Liver is the 'metabolic factory' of the body.

It has essential metabolic, synthetic and excretory functions (Table 21.1-21.3).

**Table 21.1:** Normal liver function

<b>Metabolic</b>	Carbohydrate metabolism Lipid and lipoprotein metabolism Protein metabolism
<b>Synthetic</b>	Protein synthesis: Most plasma proteins: (except Ig, RhF, complement) Blood coagulation factors: II, V, VII, IX, XI Primary bile acids Lipoprotein Urea Bilirubin glucuronide
<b>Excretory</b>	Detoxification of: Xenobiotics Drugs Toxic metabolic wastes (ammonia, bilirubin)

**Table 21.2:** Half-life of plasma proteins

<i>Plasma protein</i>	<i>t</i> <sub>1/2</sub>
Albumin	20 d
Factor VII	4–6 hrs
Transthyretin	1–2 d
Transferrin	6 d

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**Table 21.3:** Excretion and detoxification

a.	Inactivation of steroids: Cholesterol Steroid hormones
b.	Inactivation and metabolism of drugs, toxins
c.	Urea synthesis
d.	Excretion Cholesterol Bile acids Conjugated steroid hormones (with glucuronate/sulfate) Drugs Toxins

**Bilirubin metabolism** (Fig 21.1)

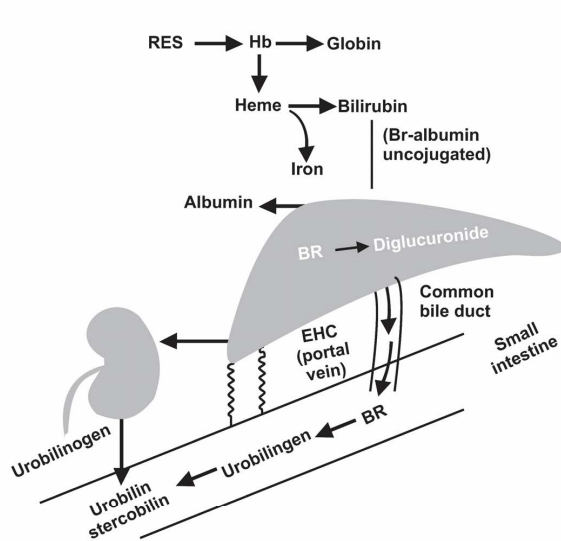


Fig. 21.1: Bilirubin metabolism

**Laboratory Tests of Bilirubin** (Fig 21.4-21.7)

- Total bilirubin
- Direct bilirubin or conjugated
- Indirect bilirubin or unconjugated (total minus direct).

**Table 21.4:** Location of hepatocellular enzymes

<i>Site</i>	<i>Enzyme</i>
Cytoplasmic	LDH AST ALT
Mitochondrial	mt AST Glutamate dehydrogenase (GDH)
Canalicular surface of hepatocyte	ALP GGT*

\* also found in microsomes

**Table 21.5:** Localization of liver damage with the help of enzymes

1. Serial enzyme determination:	ALT LDH
2. Isoenzyme determination:	CK
3. Estimation of more than one enzyme:	ALT AST

**Table 21.6:** Storage (depot) function of liver

<i>Site of storage of:</i>
Iron (700 mg iron)
Glycogen
Amino acids
Lipids (TAG)
Vitamins
<i>Investigating a liver disease:</i>
Bilirubin excretory function
Hepatocellular damage:
ALT/AST
GGT
Cholestasis:
ALP
Synthetic:
Albumin
Prothrombin time

**Table 21.7:** Tests for assessment of cell damage

Acute cell damage	Very high ALT
Cirrhosis	Slightly raised ALT
Severe liver disease (terminal)	Fall in enzyme activity

Extent of cell damage and rate of release of enzyme from damaged cells and their clearance from plasma determine the plasma levels of enzymes.

# Renal Functions

## Learning objectives:

1. Explain Renal functions and Renal function tests.
2. Describe Renal failure.
3. Illustrate Urine osmolarity and its importance clinically.
4. Discuss Acid-base balance and disorders associated with it.

## Kidneys

Kidneys are paired organ system lying in lumbar region, each having about 1 million nephrons. (Table 22.1, 22.2)

**Table 22.1:** Major functions of kidney

<i>Excretion:</i>	excretion of end products of metabolism
<i>Regulation of ecf:</i>	regulation of extracellular fluid (ecf) volume and composition and acid-base status
<i>Endocrine function:</i>	synthesis of hormones namely, renin, erythropoietin, calcitriol

## Glomerular Fluid (GF)

It is an ultrafiltrate of plasma and its composition is similar to plasma except that proteins are absent. (Table 22.3, 22.4)

## Glomerular Filtration Rate (GFR)

Normal GFR is 120 mL/min approximately and is equivalent to volume of 170 L/24 h.

*GFR depends on:*

- rate of blood flow
- balance between hydrostatic and oncotic forces in afferent arterioles and glomerular filter (tubular fluid).

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**Table 22.2:** Functions of kidney

Formation of urine  
 Regulatory function:  
     acid-base balance  
     water-electrolyte balance  
 Excretion of end products of protein metabolism  
 Endocrine function:  
     Vitamin D metabolism  
     Renin release  
     Erythropoietin production  
     PG and TX production  
 Body protein homeostasis

**Table 22.3:** Glomerular permeability: Salient features

Glomerulus is a size-, charge- and shape-dependent filter  
 Glomerular basement membrane has strong negative charge:  
 for negatively charged molecules, size cut off is 18 nm.  
 Proteins of molecular weight lower than albumin (66.3 kD) filterable.  
 Damage to negative charge results in leakage of albumin (molecular radii 3.6 nm) through basement membrane.

**Table 22.4:** Renal function tests

1. *Glomerular function:*
  - a. GFR:
    - Exogenous substance:
      - Inulin
      - Cr – EDTA
    - Endogenous substance:
      - Urea
      - Creatinine
      - $\beta$  macroglobulin
      - Retinol-binding protein
      - $\alpha$ 1 microglobulin
      - Cystatin C
  - b. *Glomerular permeability:*
    - Urine protein
2. *Tubular function:*
  - Tubular proteinuria
  - Osmolality
  - Specific gravity
  - Concentration and dilution tests
  - Tubular secretory mass measurement by diiodone/PAH

*Clearance*

It is ml of plasma cleared off a particular substance by both the kidneys per minute. (Table 22.5-22.6)

*Properties of ideal substance for clearance:*

1. It should be neither secreted nor reabsorbed by the kidney.
2. It should not be affected by rate of urine flow.
3. Its levels in blood should remain constant.
4. Its clearance should be equivalent to GFR.

**Table 22.5:** Comparison of urea and creatinine clearance

Urea clearance	Creatinine clearance
1. Its value is 3/5th of GFR	1. Values are nearly equal to GFR
2. Plasma values of urea are not constant throughout the day	2. Its plasma values remain constant over 24 hrs
3. Urea is both secreted and reabsorbed by kidney	3. Creatinine is neither secreted nor reabsorbed by kidney
4. It is affected by rate of urine flow	4. It does not depend on rate of urine flow

**Table 22.6:** Plasma urea/Creatinine ratio

Increased	Decreased
High protein intake	Low protein intake in diet
Hypercatabolic state	Severe liver disease
Dehydration	Dialysis
GI bleeding	

## Overflow Proteinuria

Abnormal amounts of low molecular weight proteins appearing in plasma and urine.

**Table 22.7:** Principal proteins that appear in urine

Protein	Disease
Amylase	Acute pancreatitis
Bence Jones	Multiple myeloma
Hemoglobin	Hemolysis (intravascular)
Lysozyme	Myeloid Leukemia
Myoglobin	Crush injury to muscle

## Microalbuminuria

More correctly termed as *minimal albuminuria*. It is mildly increased excretion (200 to 300 mg/L) of albumin in urine.

It appears to be a predictor of future development of clinical renal disease in patients with DM or hypertension.

Guidelines for investigating microalbuminuria

1. Measure albumin/creatinine, if > 30 mg/d, do albumin excretion rate (AER) in 24 hours collection.
2. AER between 30 to 300 mg/24h is microalbuminuria.
3. Further monitor AER, blood pressure, baseline GFR on every follow-up.

The incidence of microalbuminuria may be reduced by optimal control of glycemia, blood pressure.

# Pregnancy

## Learning objectives:

1. Enumerate all substances that can pass and cannot pass through placenta.
2. Describe monitoring during pregnancy.
3. Explain characteristics and features of Amniotic fluid.

## Transport Across Placenta

---

1. Passive transport
  - i. Limited passive transport:
    - FFA
    - Unconjugated steroids
    - Steroid sulfates
  - ii. Passive transport:
    - O<sub>2</sub>
    - CO<sub>2</sub>
    - Na
    - Cl
    - Urea
    - Ethanol
    - Lipid soluble molecules up to 5000 Da
2. Active transport
  - Glucose
  - Calcium
  - Many amino acids
3. *Receptor mediated endocytosis*
  - Insulin
  - LDL
  - Maternal IgG

## Substances not Transported Across Placenta

---

Maternal IgM  
Thyroid hormones  
Most proteins  
Maternal and fetal RBCs

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## MONITORING DURING PREGNANCY (Table 23.1-23.4)

**Table 23.1:** Lab evaluation during pregnancy

1st Trimester	2nd Trimester
Hb	15–21 weeks
ABO, Rh (D)	Maternal serum $\alpha$ AFP
Urine:	24–24 weeks:
Protein	Hb
Culture	Diabetes detection/screening
Toxoplasma Ab (if indicated)	Urine protein
Syphilis Ab	
HIV Ab	

**Table 23.2:** Metabolic changes during pregnancy

Decreased	Increased
Urea	Cortisol
Albumin	Copper
Total protein	Thyroxine
Glucose tolerance	HDL - C
LDL - C	ALP

**Table 23.3:** Hormonal changes during pregnancy

Hormone	Source
1. hCG	Corpus luteum
2. Estrogen	Corpus luteum Placenta Fetal Adrenal
3. Human placental lactogen	Placenta
4. Placental ALP	Placenta

**Table 23.4:** Aminocentesis: Biochemistry

Parameter	Diagnostic value
Lecithin/Sphingomyelin ratio (L/S)	Fetal lung maturity
Phosphatidylglycerol	Fetal lung maturity
Creatinine	> 2 mg/dl - Fetus is mature
Bilirubin	Reserved for monitoring blood group immunized cases
Blood acid-base and electrolyte level	Feto-placental well-being
Hormones:	Feto-placental well-being
Estriol	Feto-placental well-being
hCG	Feto-placental well-being
hPL	Feto-placental well-being
AFP	Diagnosis of neural tube defects, Down's syndrome
Chorionic villus sampling and analysis of fetal chromosome and DNA	Screening of inherited disorders

## Amniotic Fluid: Characteristics

---

### Origin \*:

Placenta  
Fetal kidney  
Skin membrane  
Lung  
Intestine

\* relative contribution of each source depends on stage of fetal development

### Biochemistry

- i. *before keratinization* of fetal skin: it can result as a *transudation* from fetal surface
- ii. *after keratinization of fetal skin*: fetal urine makes a significant contribution.

## Amniotic Fluid: Features

---

Isotonic (early pregnancy) to hypotonic (term, 255 m Osm/kg of water)

### Nitrogenous products:

Urea  
Creatinine  
Uric acid

### Proteins:

Source : fetal (skin, urinary, GIT and respiratory tract)  
: maternal (by transudation)

### Hormones:

<i>Peptide</i>	<i>Steroid</i>	<i>PG</i>
ACTH	Estradiol	PGE <sub>2</sub>
FSH	Estriol	PGF <sub>2α</sub>
LH	Estrone	
hCG		
HPL		
PRL		
Oxytocin		
Relaxin		
Thyrotropin		
Thyroxine		

## Monitoring in Gestational Diabetes\*

---

Maternal blood glucose  
OGT  
HbA<sub>1c</sub>  
Urine albumin  
Serum creatinine

\*for further details, see section on diabetes.



# **Section V**

## **Oral Cavity**

# The Oral Cavity—A Diagnostic Window to the Body

## Learning objectives:

1. Discuss saliva as a diagnostic tool.

Saliva is the protective fluid of the oral cavity with its vast supply of microbe killers, saliva combats invading pathogens such as HIV and a host of bacteria associated with oral and systemic diseases. Antibodies directed against pathogens such as polio and cold viruses, are formed in saliva. Large salivary glycoproteins called mucins, appear to have antiviral properties as well.

*Oral fluid is also a mirror of the body*, containing many compounds that are indicator of health, disease status of a person. Also, like blood and urine, composition of oral fluid may be altered in the presence of disease. Saliva may be collected in a much less invasive fashion than blood or urine, and at times preferred over blood for analyzing analytes (Table 24.1).

**Table 24.1:** Saliva as diagnostic tool

Analyte in saliva	Use
Alcohol	For alcohol screening at roadside emergency department—test cards are available
Free cortisol	Practical, convenient and accurate
Drug abuse testing	For therapeutic drug monitoring, detecting illicit drug use
Estrogen	Useful in diagnosis of and treatment of infertile women with anovulatory cycle during in vitro fertilization, monitoring hormone replacement therapy * and closely reflect free fraction of hormone
Progesterone	To investigate luteal defects in infertile female and reflect free unbound fraction
Protein	Screening IgA in infantile hypogammaglobulinemia and BMG in Sjogren's syndrome
Steroid hormones	Useful in frequent sampling
Testosterone	Indicator of androgen status

\*requires sensitive assay for accurate measurement.

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## Other Uses of Saliva Analysis

1. Diagnostic marker for early detection of cancer related protein in breast cancer.
2. Gene therapy using salivary glands offers hope in type I DM, GH deficiency and hypoparathyroidism.
3. In future, *Oral fluid chips* will be available for disease point-of-care diagnostic platform.
4. Nanotechnology based *salivary biosensor* for ultrasensitive salivary diagnostic analytes.

## Diagnostic Molecular Targets in Saliva

*Salivary transcriptome* consists of ~ 3000 mRNA that can be used for disease diagnostics. Although, salivary proteome is still several years away, the complete normal salivary transcriptome has been reported.

As a biomarker, RNA (in saliva) is as robust and as informative as any other analyte. Highly diagnostic RNA signatures have been identified for head and neck cancer and few major human systemic diseases. Latest target for candidates for salivary diagnostic are cardiovascular disease, cancer (lung, prostate, ovarian and colon), Alzheimer's disease, osteoporosis, cerebrovascular diseases, nephritis, septicemia, chronic respiratory diseases, chronic liver disease and pneumonitis.

### Questions:

1. How is saliva used as diagnostic tool?
2. Explain diagnostic Molecular Targets in saliva.
3. What are the uses of saliva analysis?

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